

# Aqueous activity on chondrite parent asteroids

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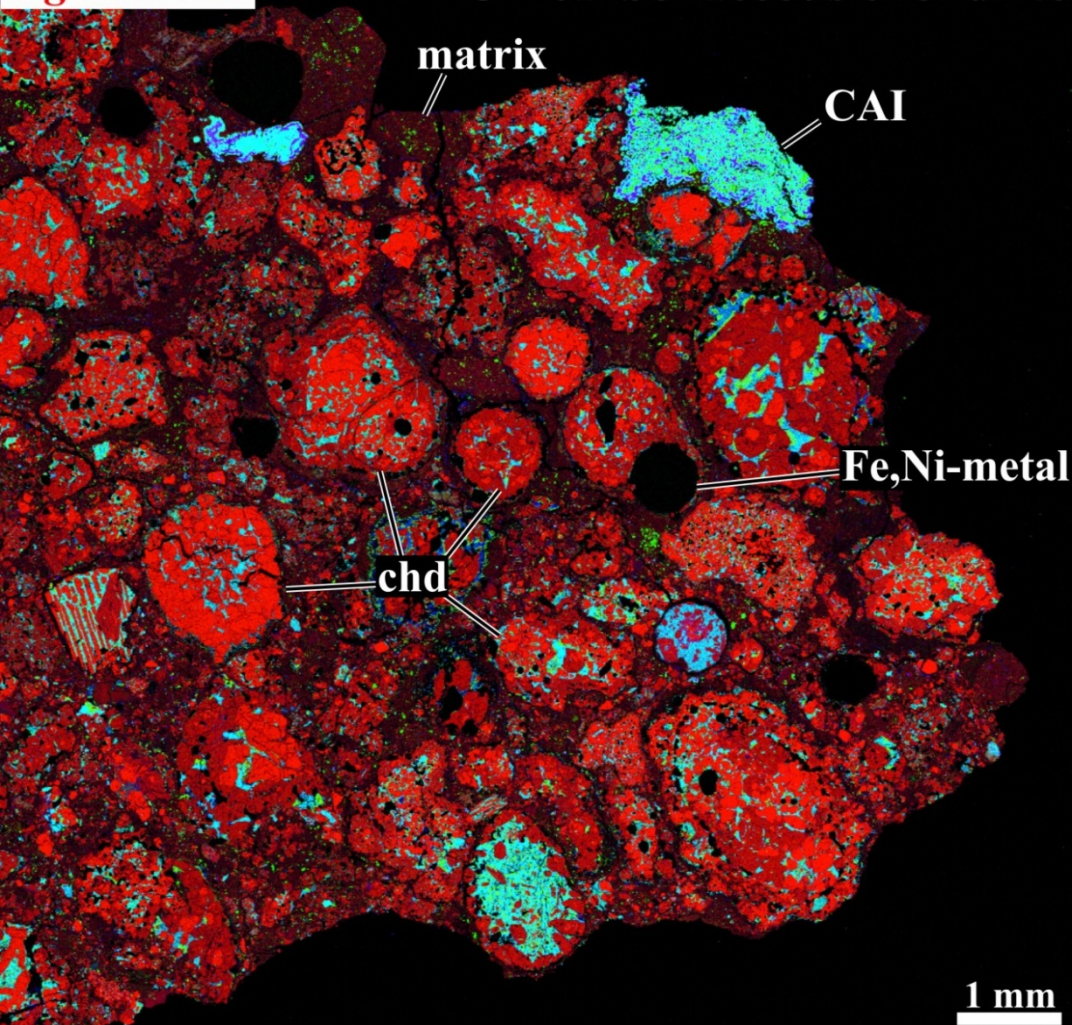
# Outline

- Chondritic meteorites & their classification
- Aqueously-formed minerals & conditions (temperature & water/rock ratio) of chondrite aqueous alteration
- Dating of aqueous alteration
  - $^{53}\text{Mn}$ - $^{53}\text{Cr}$  ages of fayalite in CV, CO, & ordinary chondrites
  - $^{53}\text{Mn}$ - $^{53}\text{Cr}$  ages of carbonates in CI, CM, & CR chondrites
- Oxygen-isotope compositions of aqueously-formed minerals & estimated D/H values of asteroidal water: Constraints on the origin of asteroidal water



# Major chondritic components: Chondrules, matrix, & CAIs

Mg : Ca : Al



- chondrules + Fe,Ni-metal (30–98 vol%)
- Ca,Al-rich inclusions (CAIs) (<1–5 vol%)
- matrix (<2–70 vol%)

In contrast to ISM, chondrites are dominated by crystalline material  
→ thermal processing in PPD

# Classification of chondritic meteorites (chondrites)

- based on mineralogy, petrography, bulk oxygen-isotope & chemical compositions, chondrites are divided into 15 groups & 3 major classes

<u>Carbonaceous</u>								<u>Enstatite</u>		<u>Ordinary</u>			Other	
CI	CM	CR	CV	CK	CO	CB	CH	EH	EL	H	L	LL	K	R
1	2-3	2-3	3	3-6	3	3	3	3-6		3-6			3	3-6

- letters designating groups refer to a typical meteorite in a group:

CI – Ivuna type

CM – Mighei type

CR – Renazzo type

CV – Vigarano type

CK – Karoonda type

CO – Ornans type

CB – Bencubbin type

- some chondrites are ungrouped (e.g., Tagish Lake)

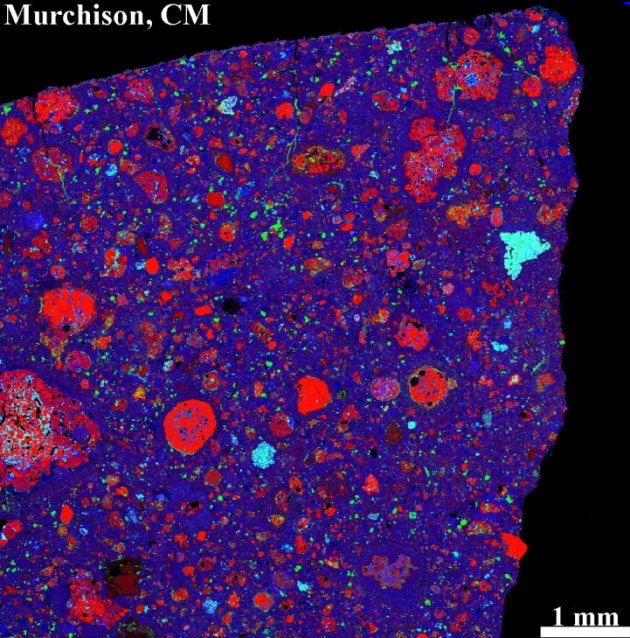
- petrologic types: 1            2            3            4            5            6

aqueous alteration

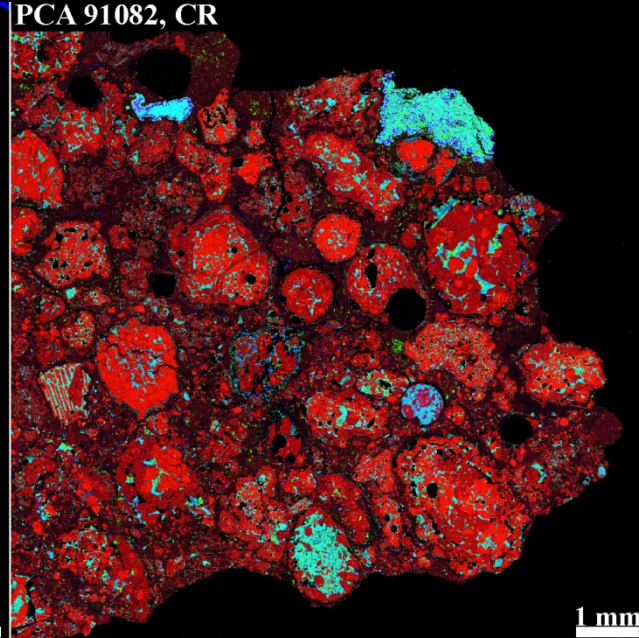
thermal metamorphism



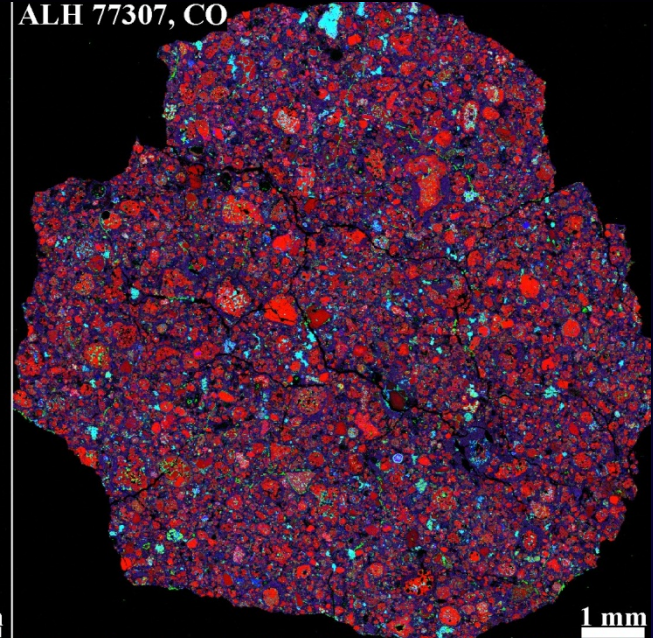
Murchison, CM



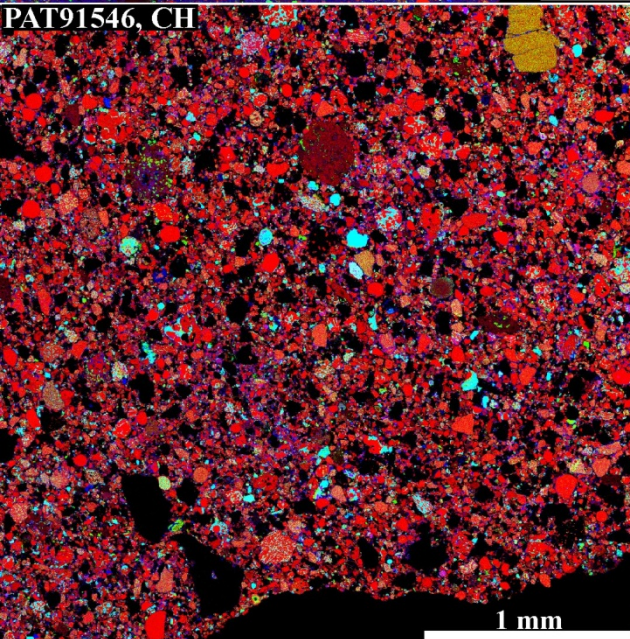
PCA 91082, CR



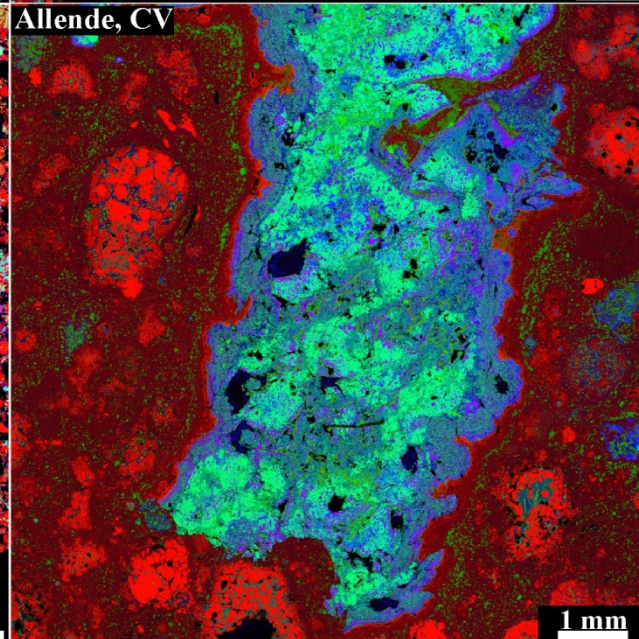
ALH 77307, CO



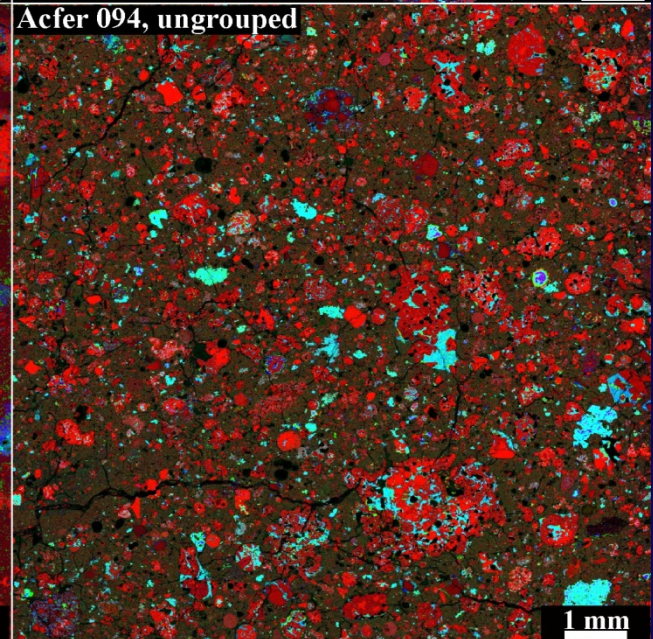
PAT91546, CH



Allende, CV

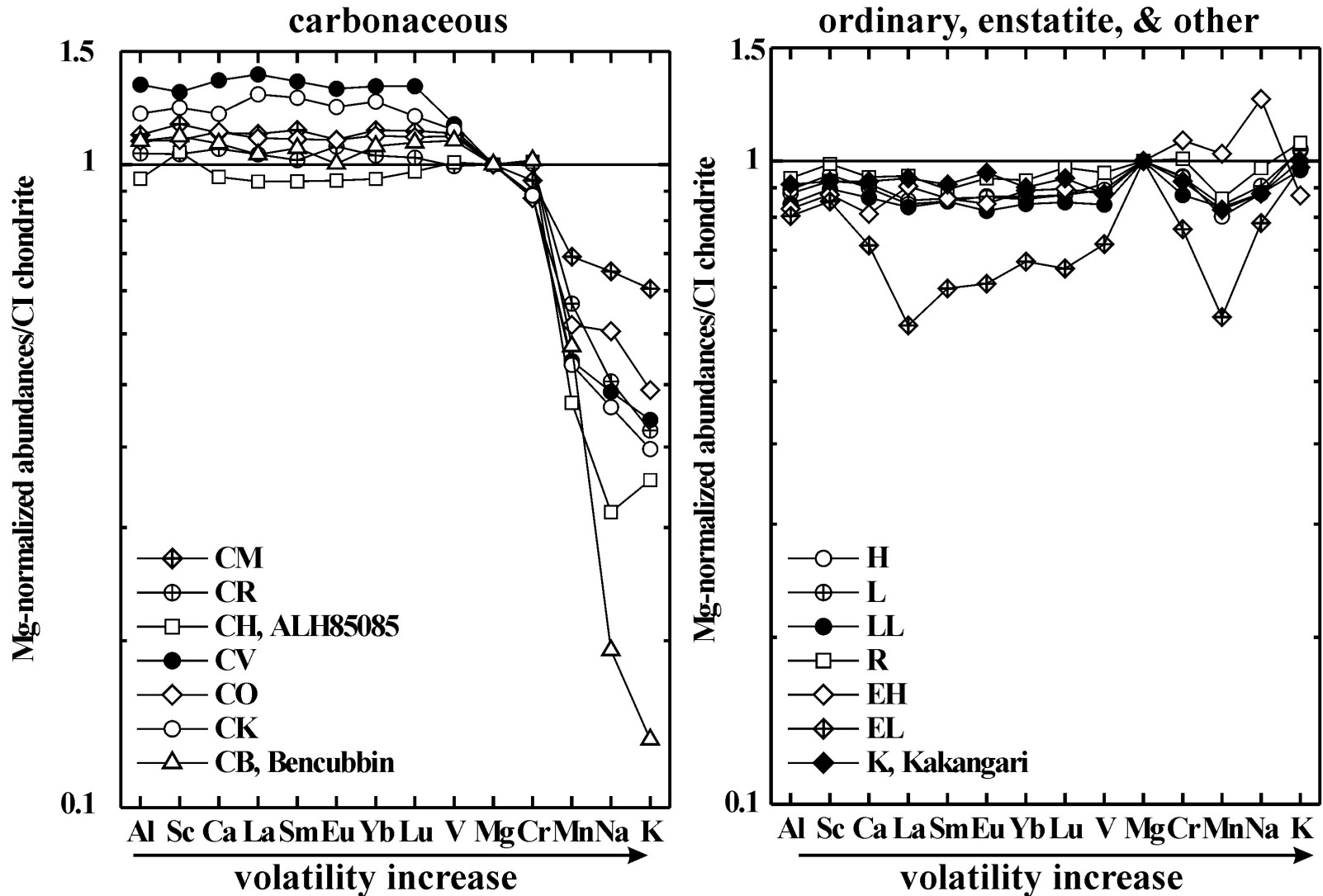


Acfer 094, ungrouped



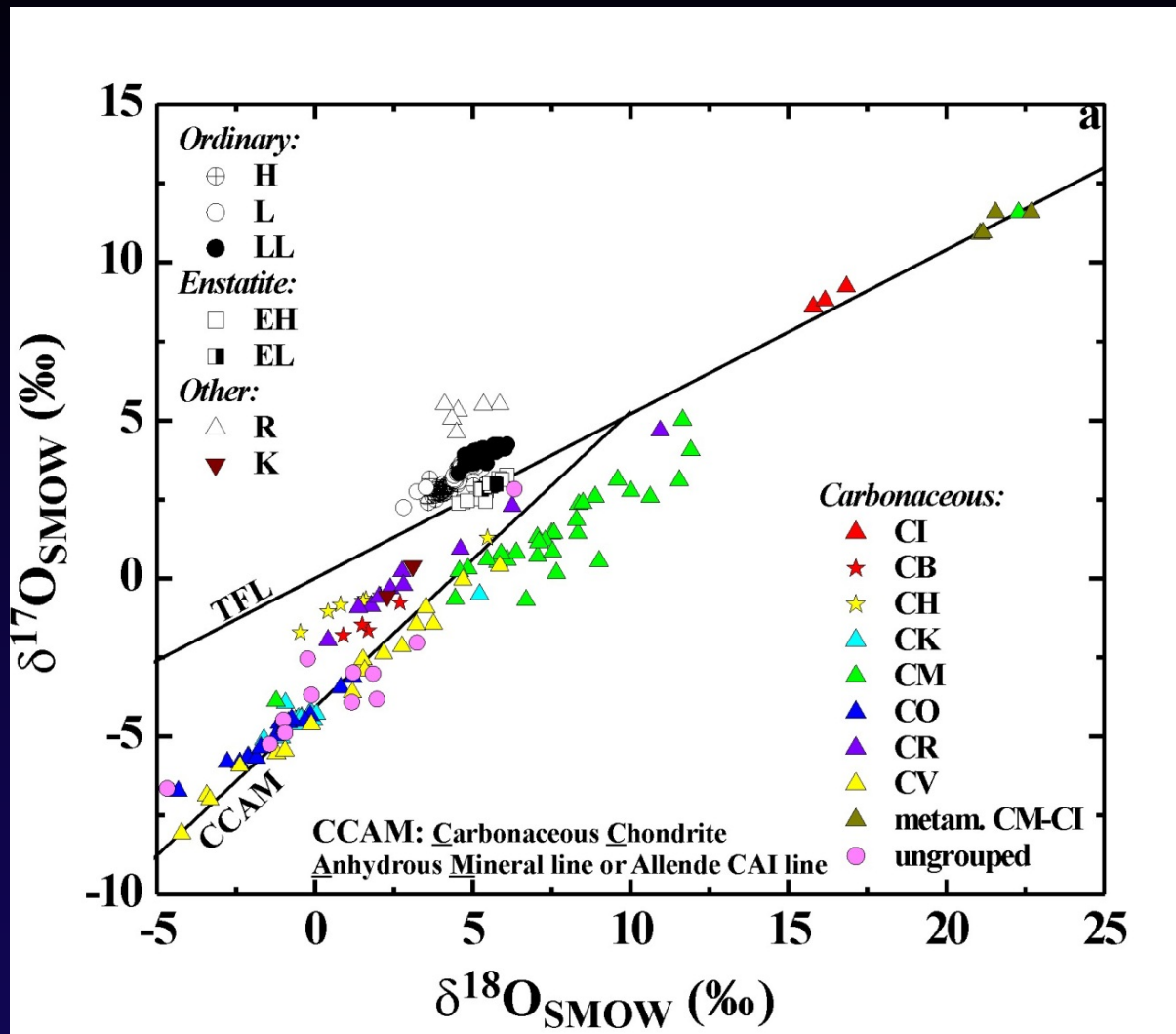
• variations in mineralogy, sizes & abundances of chondritic components among the groups

# Bulk chemical compositions



- chondrite groups have distinct bulk chemical compositions

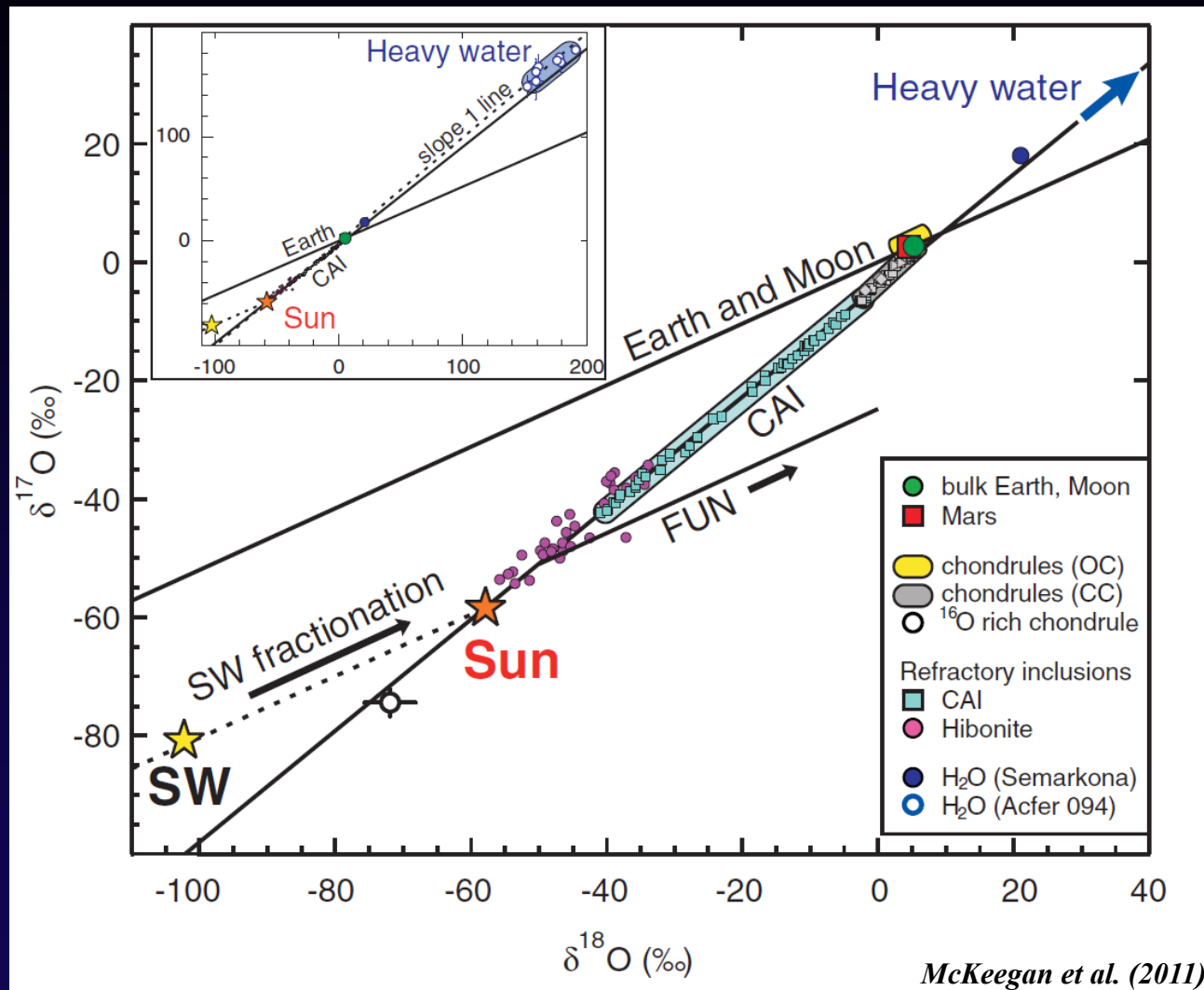
# Bulk oxygen-isotope compositions



$$\delta^{17,18}\text{O} = \left[ \frac{(^{17,18}\text{O}/^{16}\text{O})_{\text{sample}}}{(^{17,18}\text{O}/^{16}\text{O})_{\text{SMOW}}} - 1 \right] \times 1000, \text{ Standard Mean Ocean Water}$$

- chondrite groups occupy distinct regions on three-oxygen isotope diagram
- chondrite O-isotope compositions very different from Sun,  $\delta^{17,18}\text{O} \approx -50\text{‰}$

# Bulk oxygen-isotope compositions



$$\delta^{17,18}\text{O} = \left[ \frac{(^{17,18}\text{O}/^{16}\text{O})_{\text{sample}}}{(^{17,18}\text{O}/^{16}\text{O})_{\text{SMOW}}} - 1 \right] \times 1000, \text{ Standard Mean Ocean Water}$$

- chondrite groups occupy distinct regions on three-oxygen isotope diagram
- chondrite O-isotope compositions very different from Sun,  $\delta^{17,18}\text{O} \approx -60\text{‰}$



# Aqueously-altered chondrites

*Carbonaceous*

*Enstatite*

*Ordinary*

*Other*

*Ungrouped*

CI CM CO CV CK CR CH CB

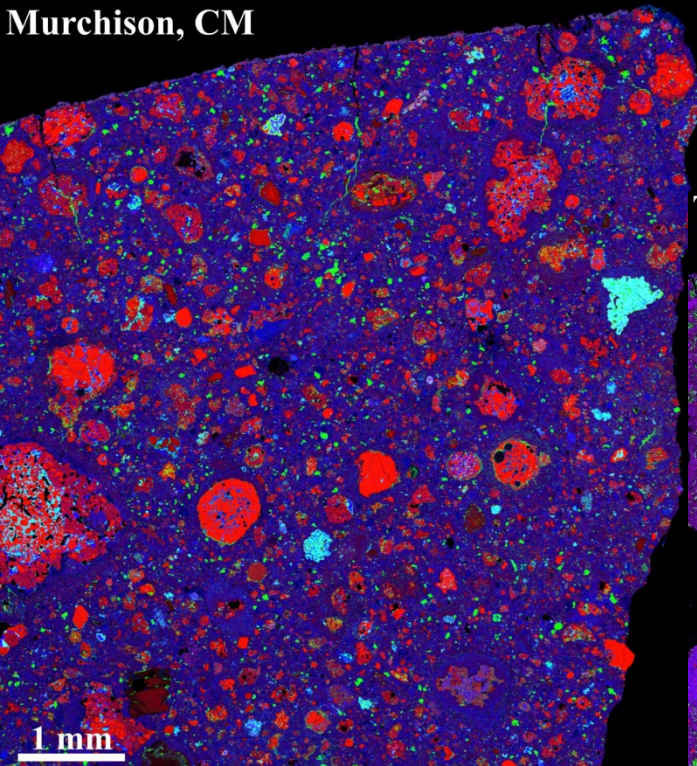
EH EL

H L LL

K R

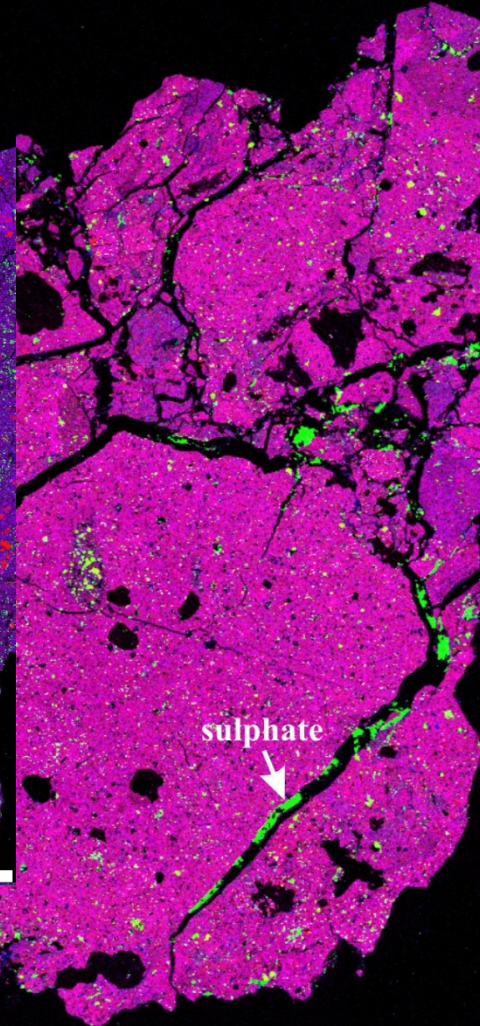
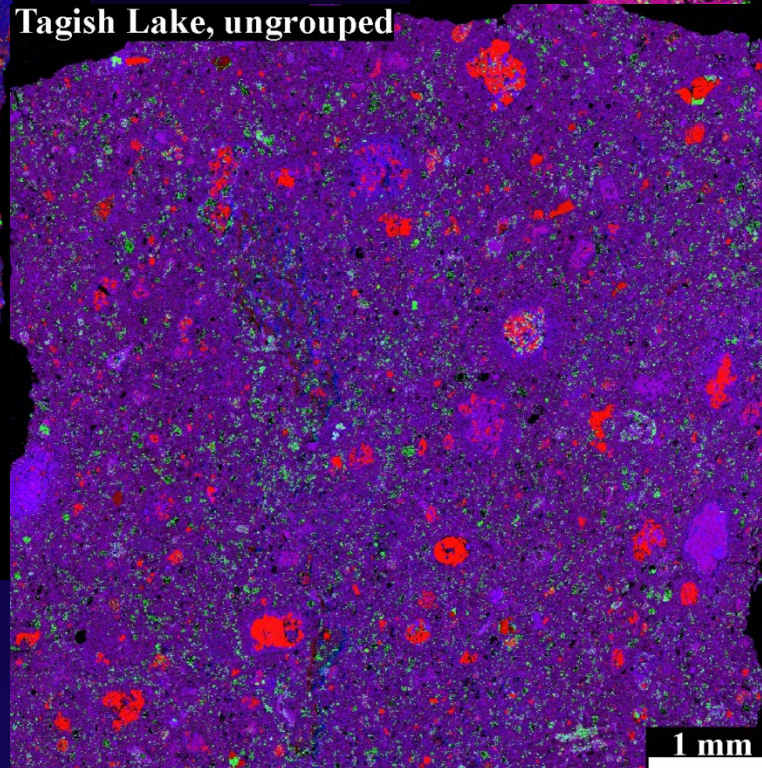
Tagish Lake

Murchison, CM



Orgueil, CI

Tagish Lake, ungrouped



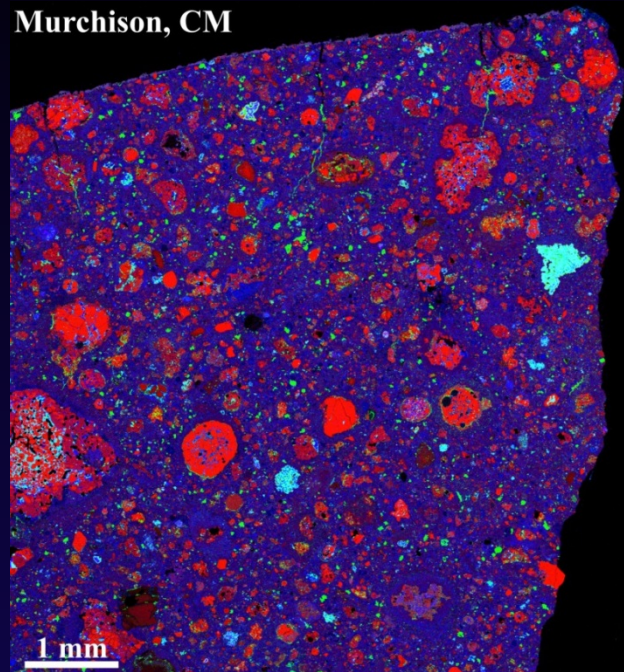
# Aqueously produced minerals in CI, CM, & CR chondrites

- CM, CI & CR chondrites experienced low-T (< 100°C) aqueous alteration at high water/rock (W/R) ratio (0.2–1)
- **phyllosilicates**
  - serpentine  $\text{Mg}_3\text{Si}_2\text{O}_5(\text{OH})_4$
  - cronstedtite  $\text{Fe}_2^{2+}\text{Fe}^{3+}(\text{Si},\text{Fe}^{3+})_2\text{O}_5(\text{OH})_4$
- tochilinite  $2[\text{Fe},\text{Mg},\text{Cu},\text{Ni}(\text{S})]\text{S}\cdot 1.57\text{--}1.85[(\text{Mg},\text{Fe},\text{Ni},\text{Al},\text{Ca})(\text{OH})_2]$
- **carbonates**
  - calcite  $\text{CaCO}_3$  (trigonal)
  - dolomite  $(\text{Ca},\text{Mg},\text{Fe},\text{Mn})\text{CO}_3$
  - breunnerite  $\text{Mg}(\text{Fe},\text{Mn})(\text{CO}_3)_2$
  - aragonite  $\text{CaCO}_3$  (hexagonal)
- **sulfides**
  - troilite  $\text{FeS}$
  - pyrrhotite  $\text{Fe}_{1-x}\text{S}$
  - pentlandite  $(\text{Fe},\text{Ni})_9\text{S}_8$
- **magnetite**  $\text{Fe}_3\text{O}_4$
- sulfates  $(\text{Mg},\text{Na},\text{Ca},\text{Ni})\text{SO}_4\cdot n\text{H}_2\text{O}$
- cohenite  $(\text{Fe},\text{Ni})_3\text{C}$



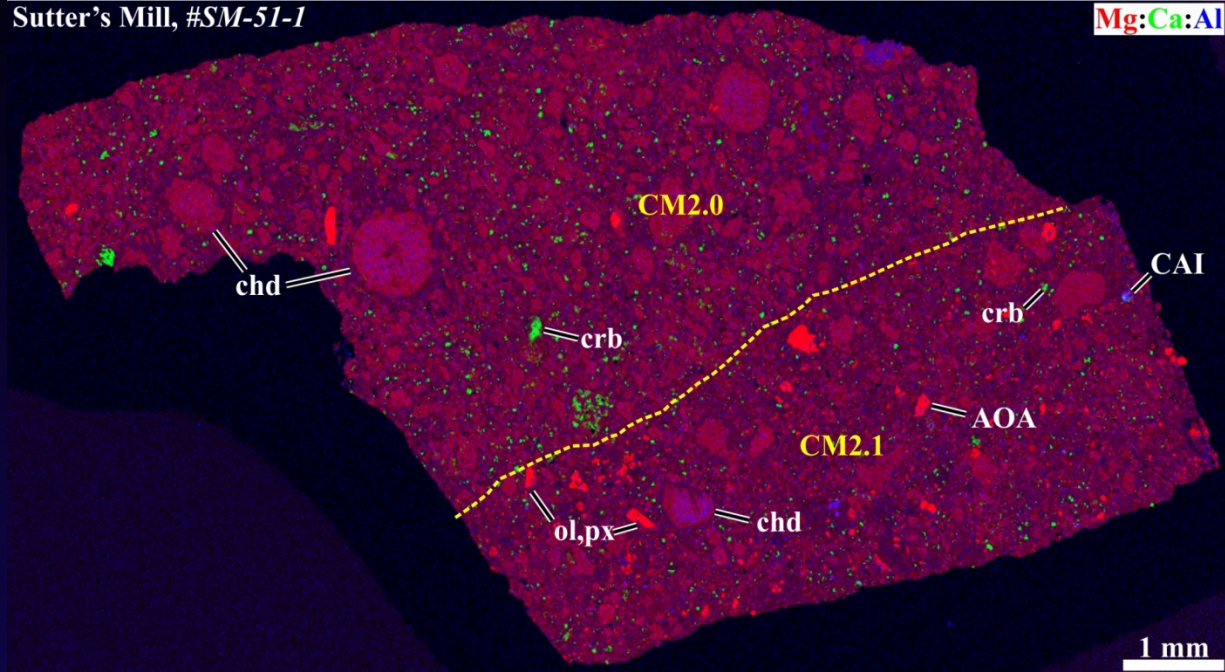
# Mineralogy & petrography of Sutter's Mill (CM)

Murchison, CM

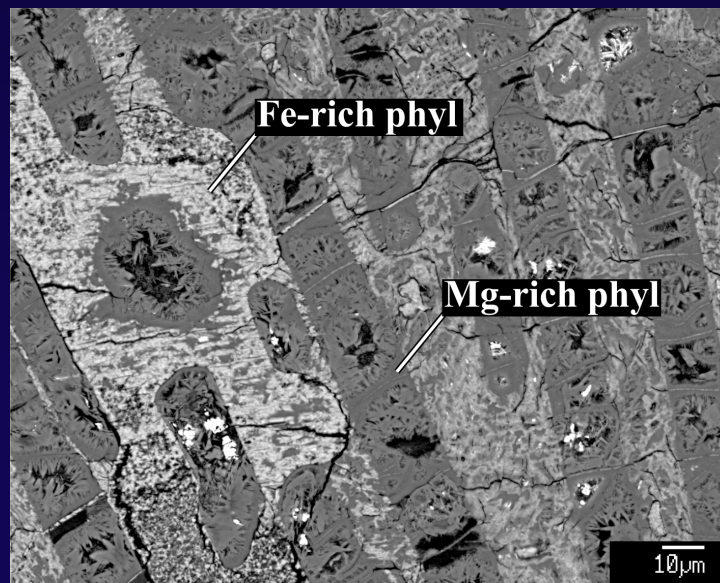
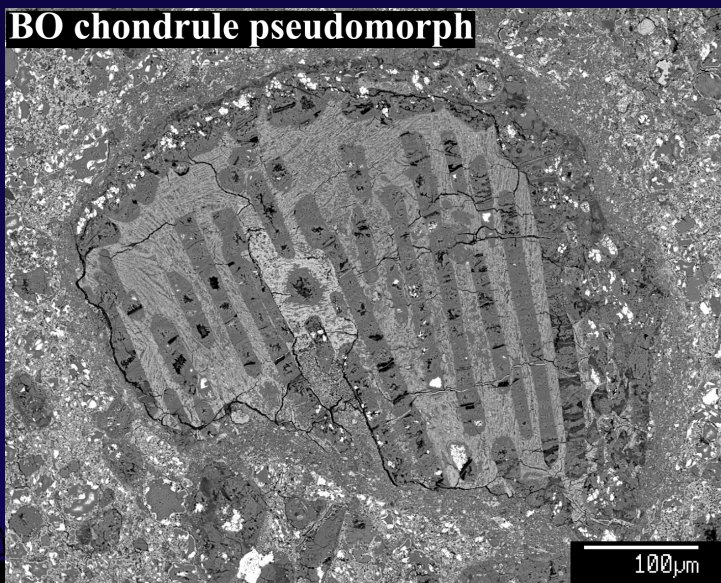


Sutter's Mill, #SM-51-1

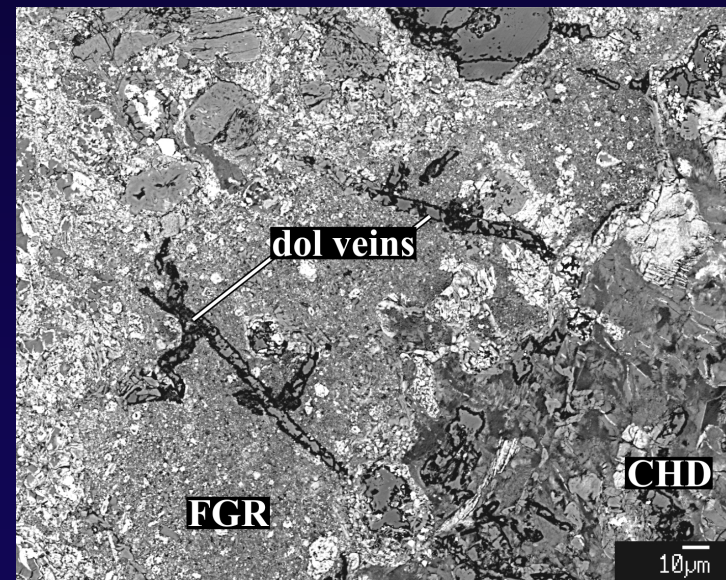
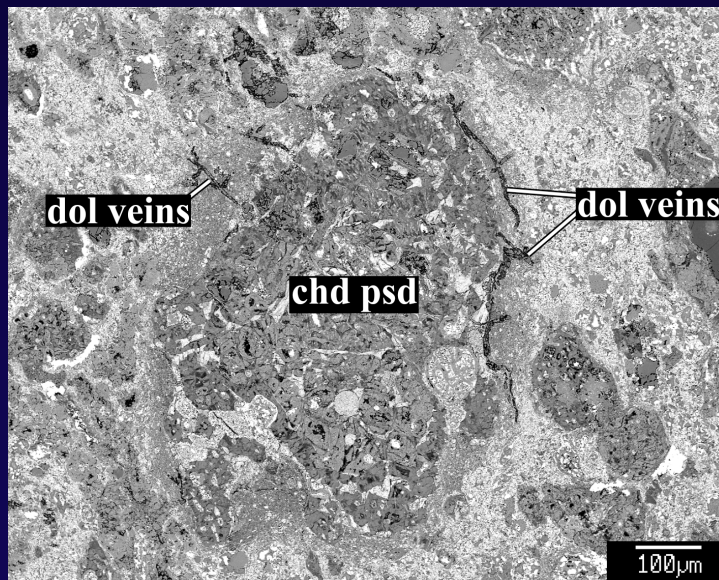
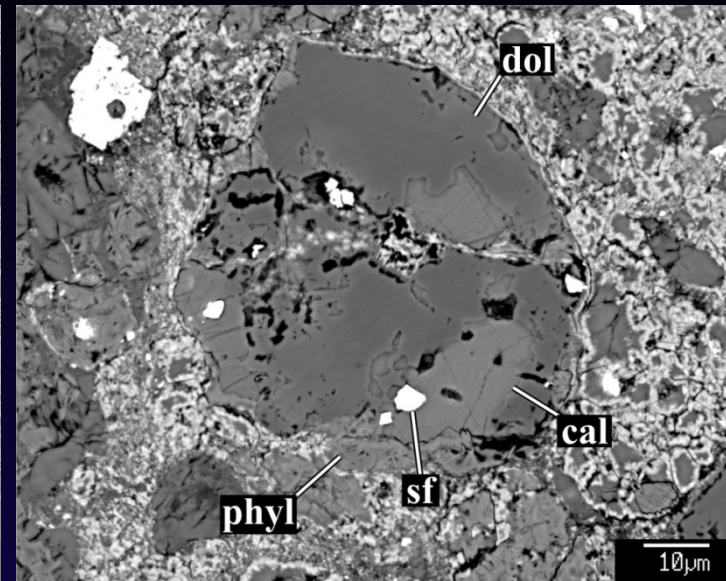
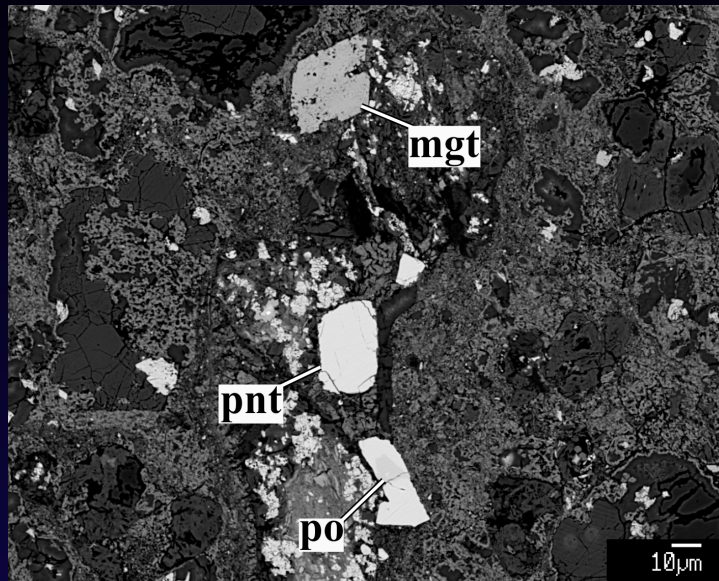
Mg:Ca:Al



**BO chondrule pseudomorph**



# Mineralogy & petrography of Sutter's Mill (CM)

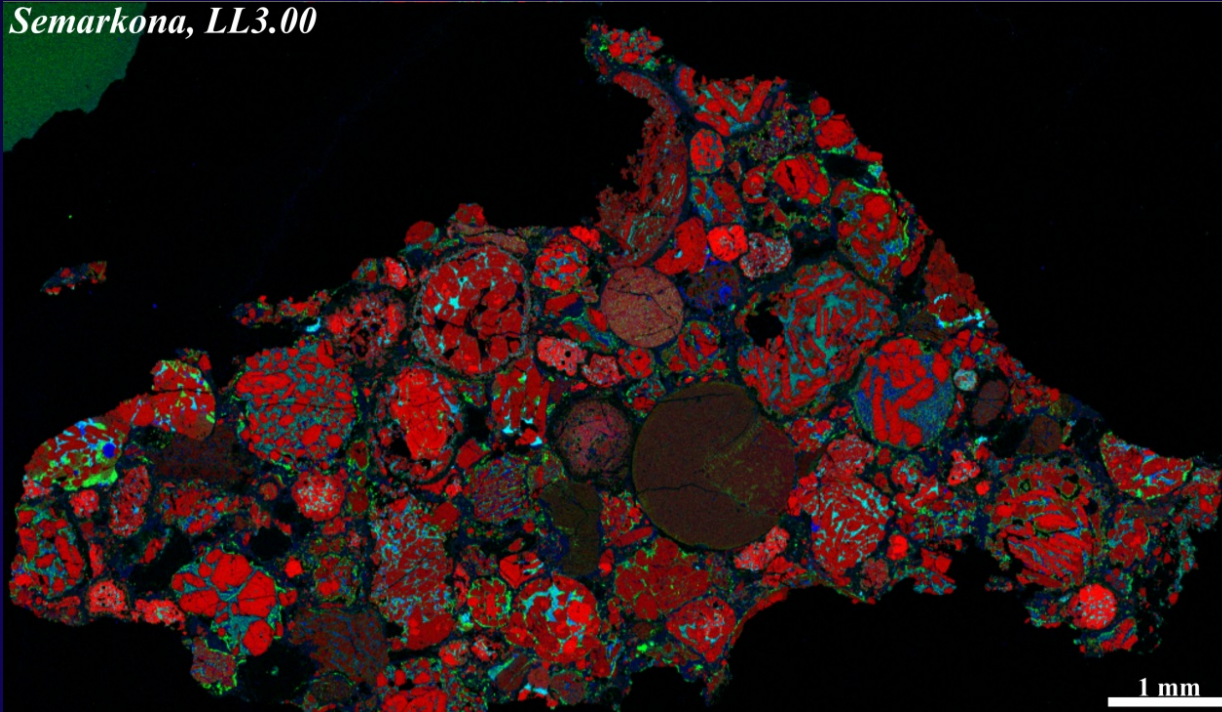


- pseudomorphic replacement of chondrules by phyllosilicates & presence of carbonate veins are clear evidence for *in situ* aqueous alteration

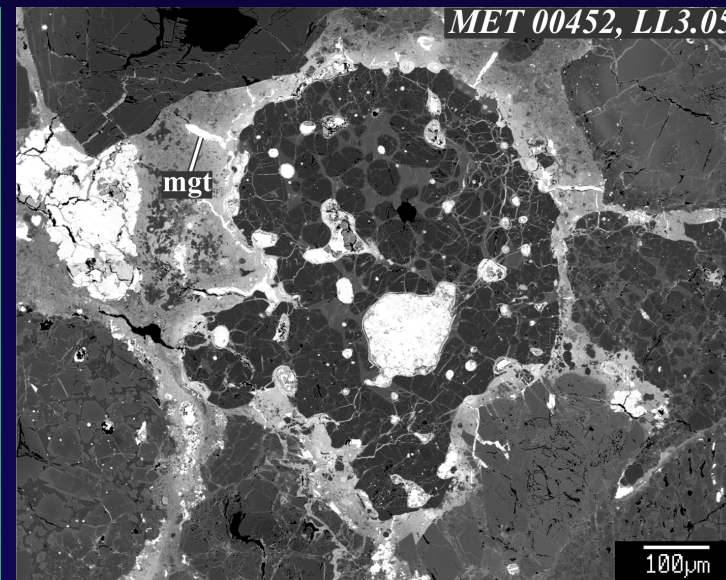
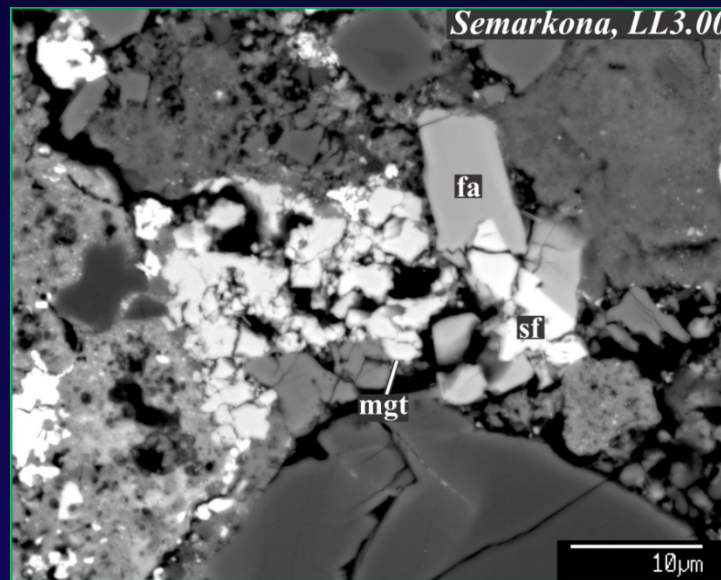
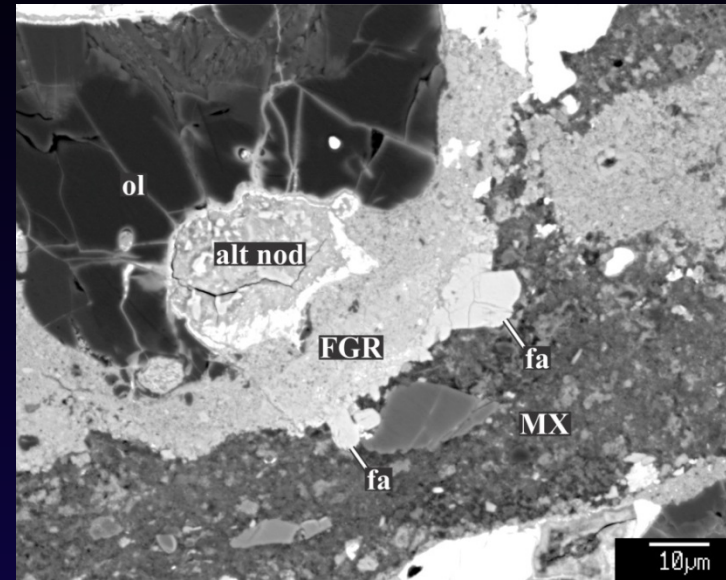
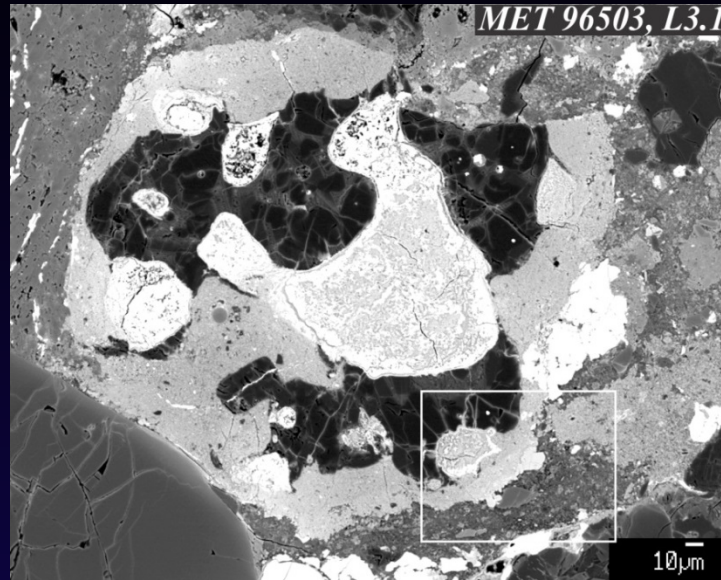
# Aqueously produced minerals in ordinary chondrites

- ordinary chondrites experienced aqueous alteration at  $\sim 100\text{--}200^\circ\text{C}$  & low W/R ratio ( $<0.2$ ) & were subsequently metamorphosed at higher temperature
- **saponite**  $\text{Ca}_{0.25}(\text{Mg,Fe})_3(\text{Si}_4\text{O}_{10})(\text{OH})_2 \cdot n\text{H}_2\text{O}$
- **magnetite**  $\text{Fe}_3\text{O}_4$
- **fayalite**  $\text{Fe}_2\text{SiO}_4$
- Fe,Ni-carbides  $(\text{Fe,Ni})_3\text{C}$  ,  $(\text{Fe,Ni})_{23}\text{C}_6$
- pentlandite  $(\text{Fe,Ni})_9\text{S}_8$
- Ni-rich metal  $\text{FeNi}$ ,  $\text{FeNi}_3$
- nepheline  $\text{NaAlSiO}_4$

*Semarkona, LL3.00*



# Aqueously produced minerals in ordinary chondrites



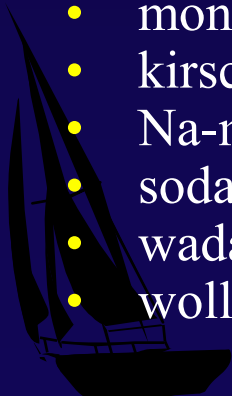
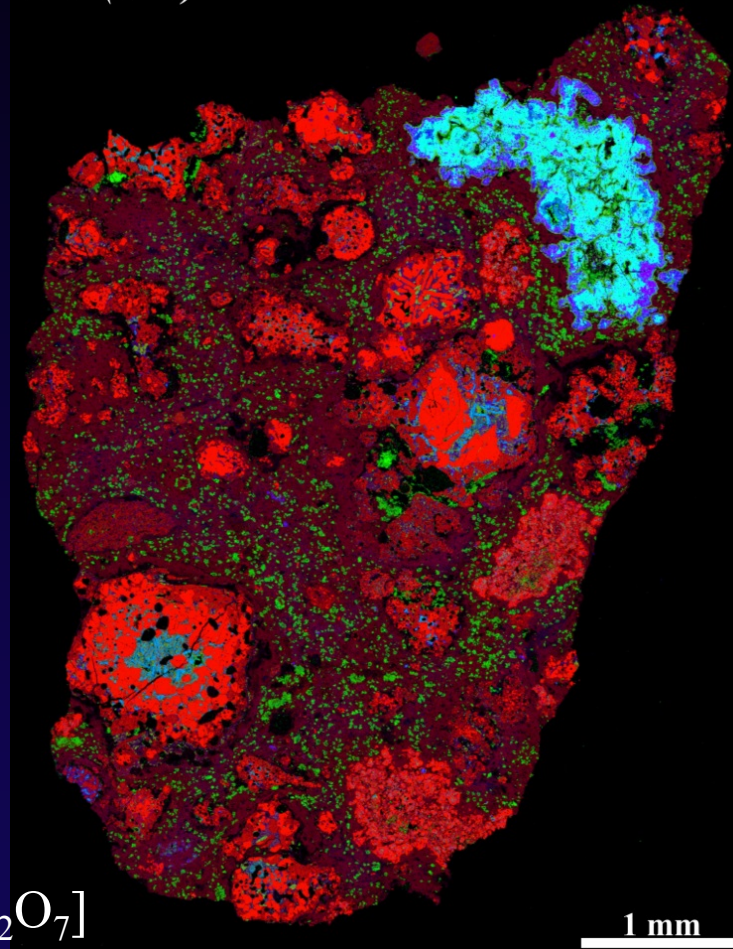
- *in situ* growth of fayalite & magnetite

# Aqueously produced minerals in CV & CO chondrites

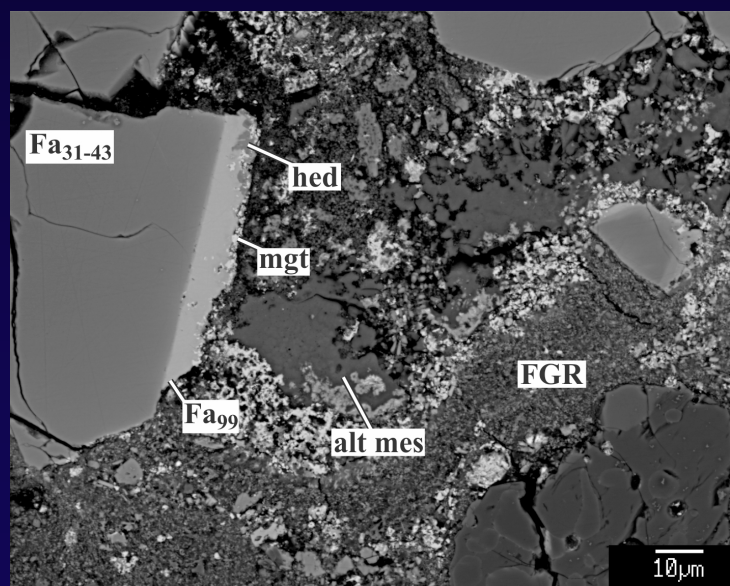
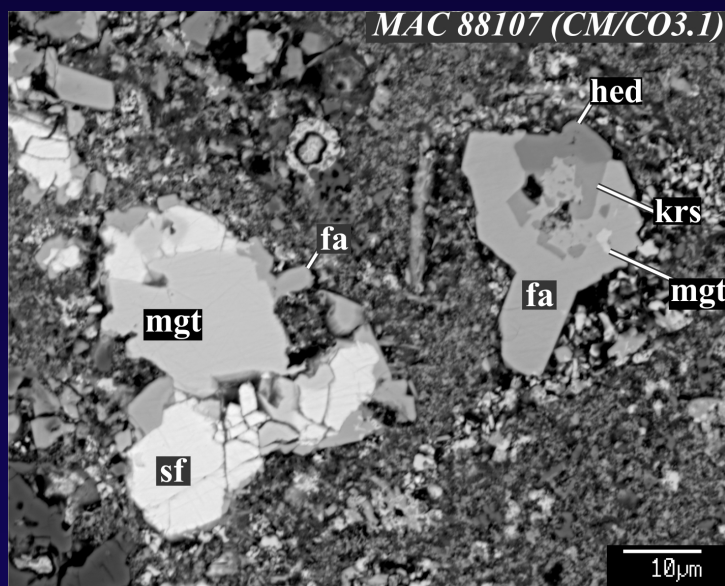
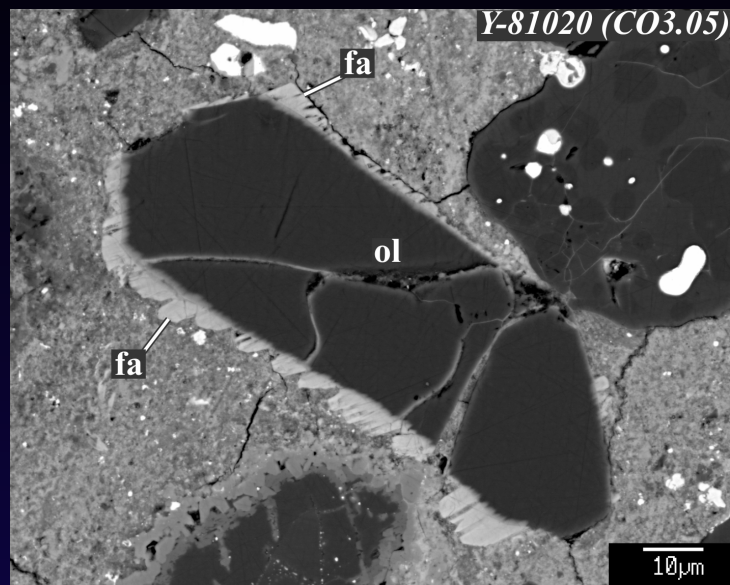
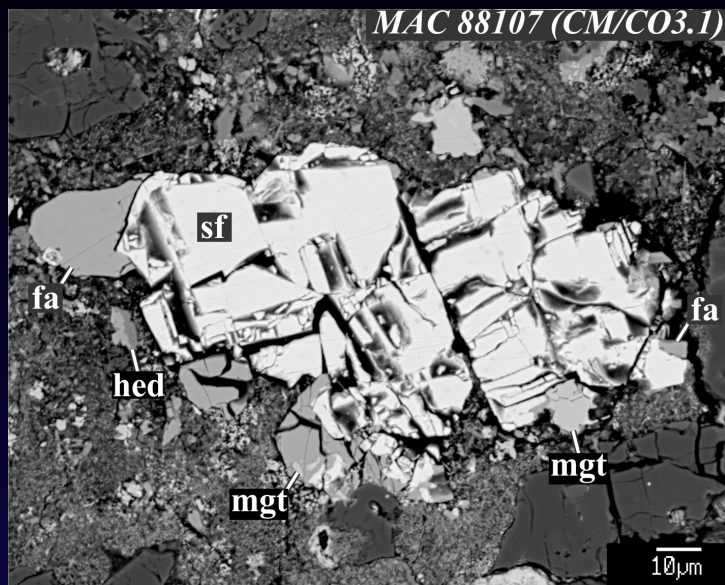
- CV & CO chondrites experienced aqueous alteration ~100–200°C & low W/R ratio (<0.2) & were subsequently metamorphosed at higher temperature

- fayalite  $\text{Fe}_2\text{SiO}_4$
- magnetite  $\text{Fe}_3\text{O}_4$
- phyllosilicates
- pentlandite  $(\text{Fe,Ni})_9\text{S}_8$
- hedenbergite  $\text{CaFeSi}_2\text{O}_6$
- andradite  $\text{Ca}_3\text{Fe}_2\text{Si}_3\text{O}_{12}$
- Fe,Ni-carbides  $(\text{Fe,Ni})_3\text{C}$ ,  $(\text{Fe,Ni})_{23}\text{C}_6$
- Ni-rich metal  $\text{FeNi}$ ,  $\text{FeNi}_3$
- nepheline  $\text{NaAlSi}_3\text{O}_8$
- anorthite  $\text{CaAl}_2\text{Si}_2\text{O}_8$
- corundum  $\text{Al}_2\text{O}_3$
- forsterite  $\text{Mg}_2\text{SiO}_4$
- grossular  $\text{Ca}_3\text{Al}_2\text{Si}_3\text{O}_{12}$
- kushiroite  $\text{CaAl}_2\text{SiO}_6$
- monticellite  $\text{CaMgSiO}_4$
- kirschsteinite  $\text{CaFeSiO}_4$
- Na-melilite  $(\text{CaNa})_2(\text{Al,Mg})[(\text{AlSi})_2\text{O}_7]$
- sodalite  $\text{Na}_8\text{Al}_6\text{Si}_6\text{O}_{24}\text{Cl}_{12}$
- wadalite  $\text{Ca}_6(\text{Al,Si,Mg})_7\text{O}_{16}\text{Cl}_3$
- wollastonite  $\text{CaSiO}_3$

*Kaba (CV3)*



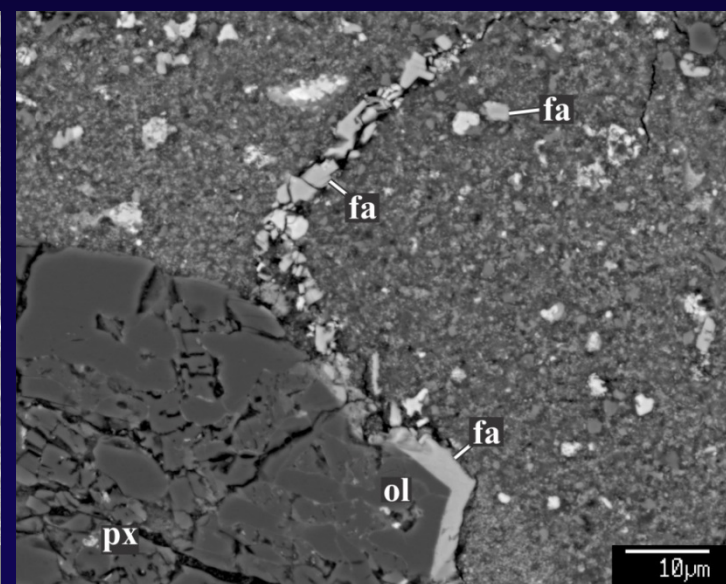
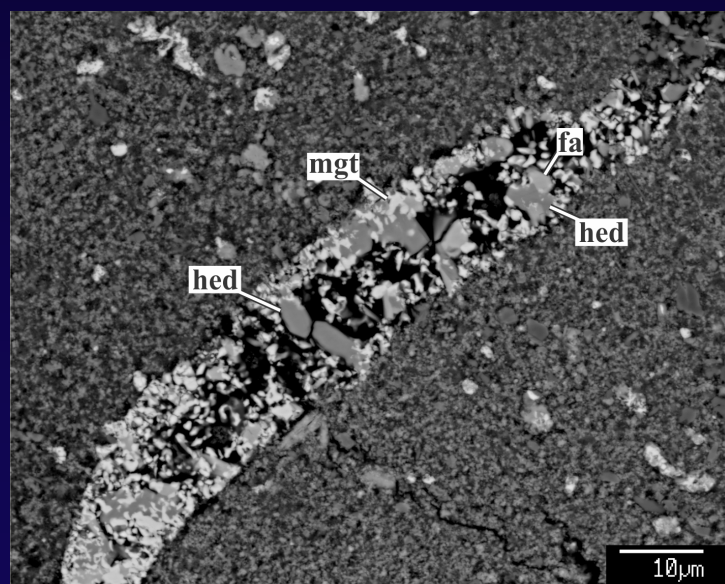
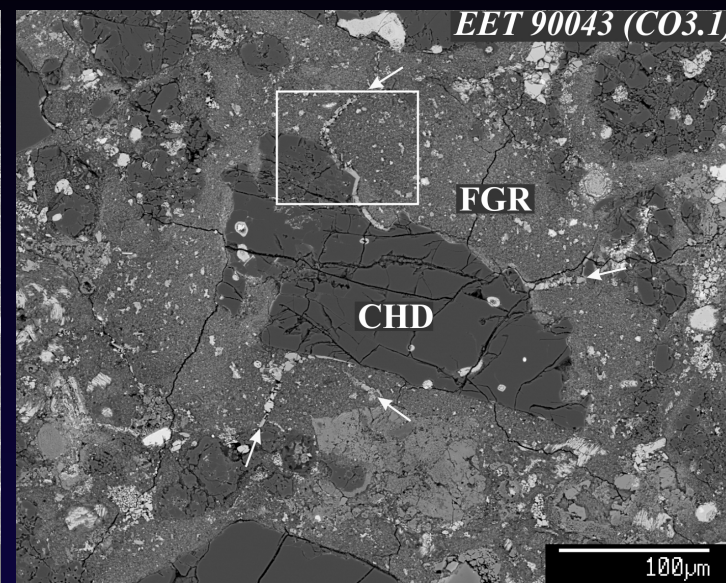
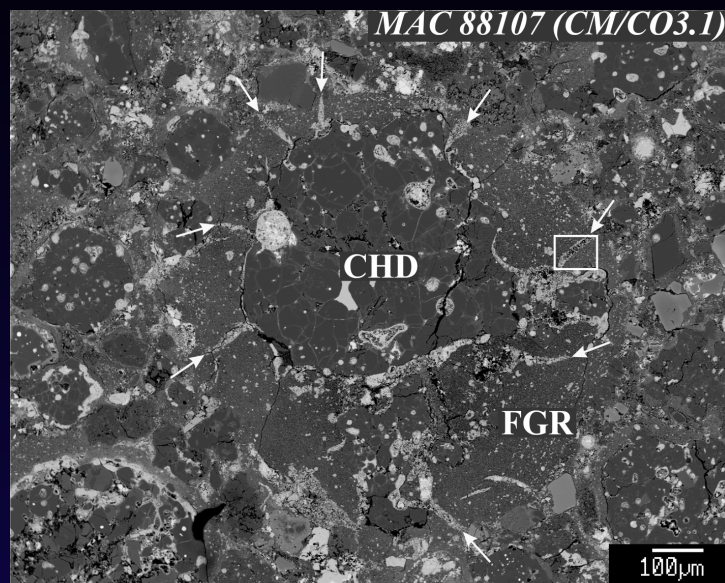
# Aqueously produced minerals in CV & CO chondrites



- fayalite associated with magnetite, hedenbergite, kirschsteinite, & Fe,Ni-sulfides in hydrated matrices; no corrosion by phyllosilicates

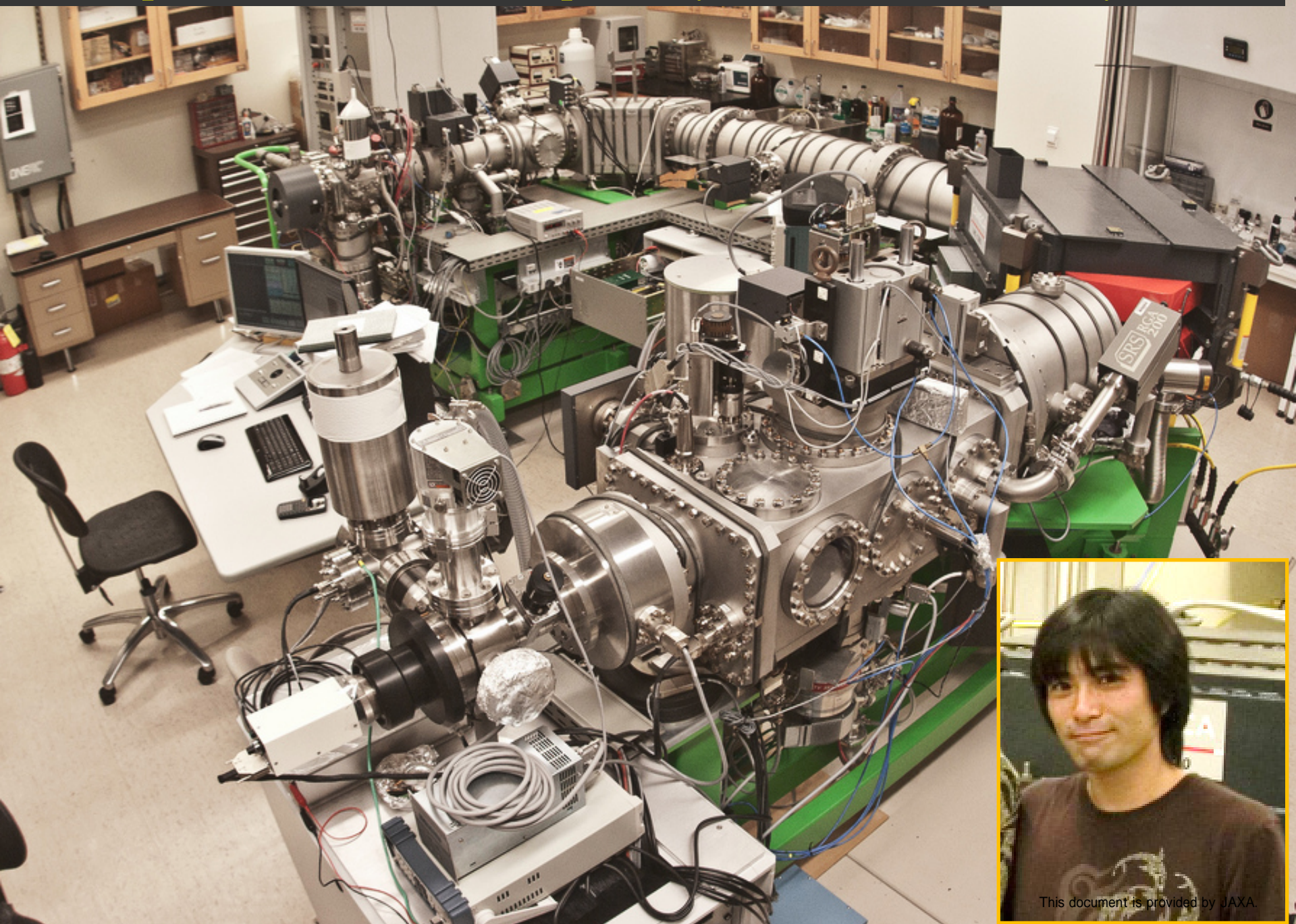


# Aqueously produced minerals in CV & CO chondrites

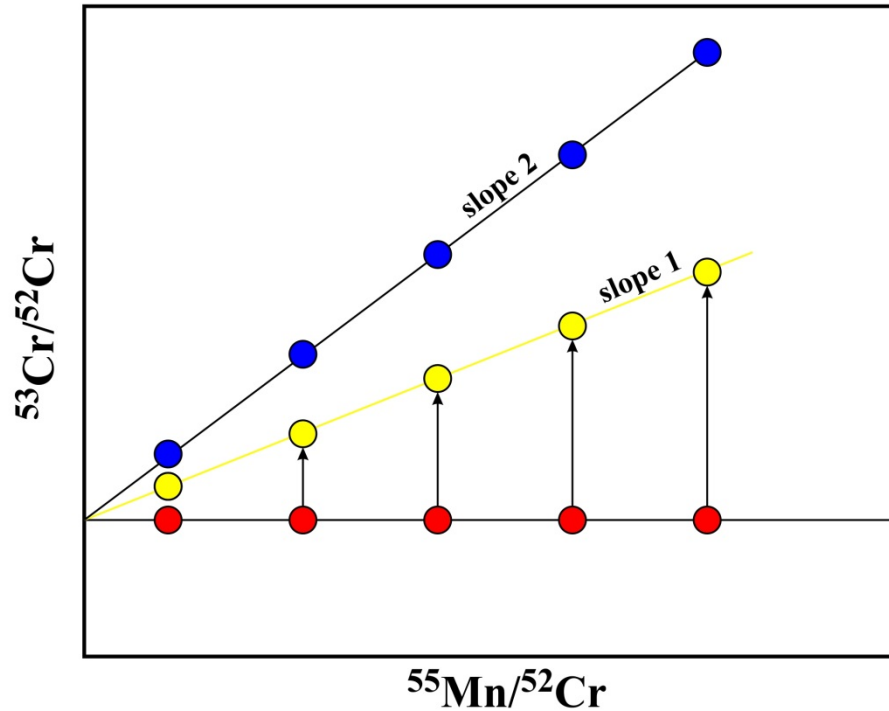


- fayalite-hedenbergite-magnetite veins, indicative for *in situ* growth

# Isotope measurements of aqueously-altered meteorites by SIMS



# Dating of aqueous alteration using Mn-Cr isotope system



- $^{53}\text{Mn} \rightarrow ^{53}\text{Cr}$ ,  $t_{1/2} \sim 3.7$  Myr
- $^{53}\text{Mn}$  was present & uniformly distributed in the early Solar System
- can be used for chronology of the early Solar System processes, including aqueous alteration

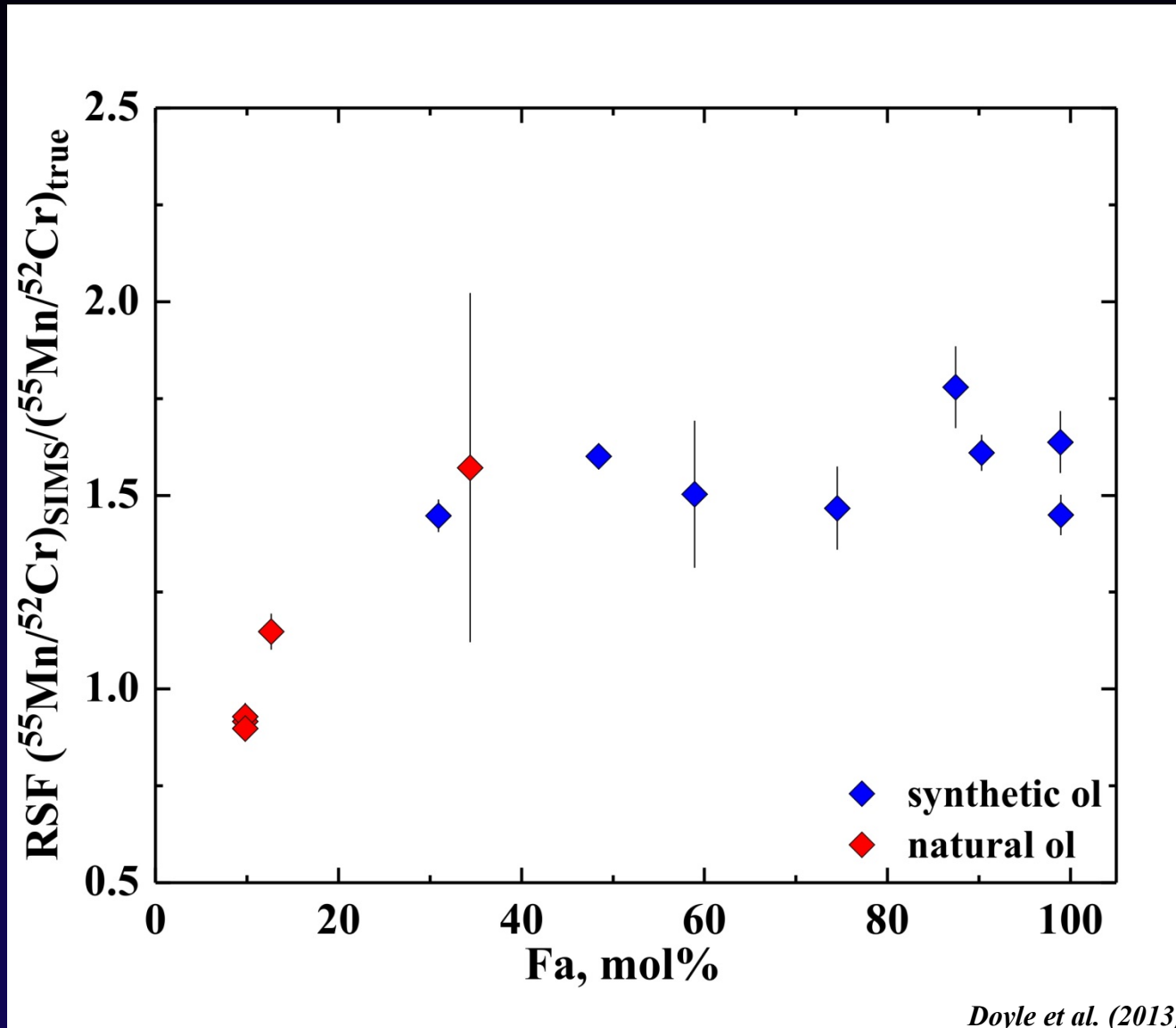
- $^{53}\text{Mn}$ - $^{53}\text{Cr}$  is a relative chronometer
- $\Delta t_{\text{slope-1} - \text{slope-2}} = 1/\lambda \times \ln[(^{53}\text{Mn}/^{55}\text{Mn})_{\text{slope-1}} / (^{53}\text{Mn}/^{55}\text{Mn})_{\text{slope-2}}]$ , where  $\lambda$  is a decay constant of  $^{53}\text{Mn}$  ( $= \ln(2)/t_{1/2}$ )
- $^{53}\text{Mn}$ - $^{53}\text{Cr}$  chronology can be anchored to absolute U-Pb chronology, e.g., D'Orbigny angrite with  $(^{53}\text{Mn}/^{55}\text{Mn})_0 = (3.24 \pm 0.04) \times 10^{-6}$  & U-corrected Pb-Pb age of  $4563.4 \pm 0.3$  Ma (Glavin et al. 2008; Brennecka et al. 2012) & compared with U-corrected Pb-Pb ages of CV CAIs  $4567.30 \pm 0.16$  Ma (Connelly et al. 2012)

# Dating aqueous alteration: $^{53}\text{Mn}$ - $^{53}\text{Cr}$ ages of carbonates & fayalite

- Relative Sensitivity Factor (RSF) =  $(^{55}\text{Mn}^+ / ^{53}\text{Cr}^+)_{\text{SIMS}} / (^{55}\text{Mn} / ^{53}\text{Cr})_{\text{EPMA}}$ ; it depends on a number of factors (*Hoppe et al. 2007; Sugiura et al. 2010; McKibbin et al. 2013; Doyle et al. 2013*)
  - mineral composition
  - analytical conditions
- proper standards for Mn-Cr isotope measurements of carbonates & fayalite are required
- San Carlos olivine ( $\text{Fa}_{10}$ ) was used a standard for  $\text{Fa}_{95-100}$  & carbonates
  - Mn- & Cr-bearing calcite (*Sugiura et al. 2010*)
  - Mn- & Cr-bearing fayalite (*Doyle et al. 2013*)

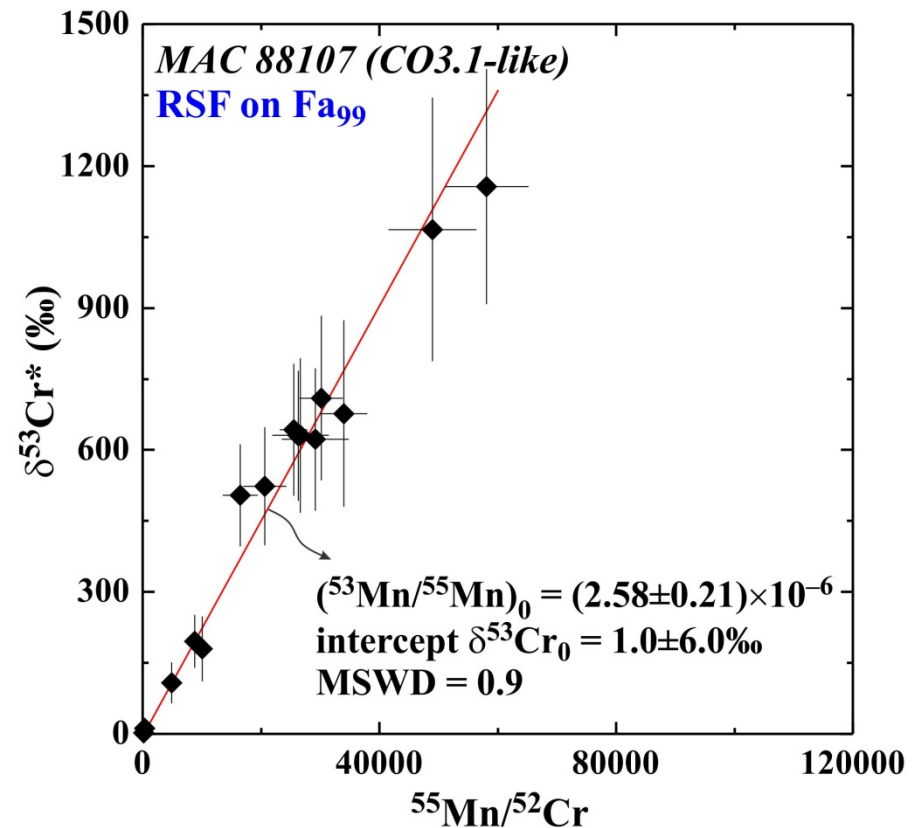
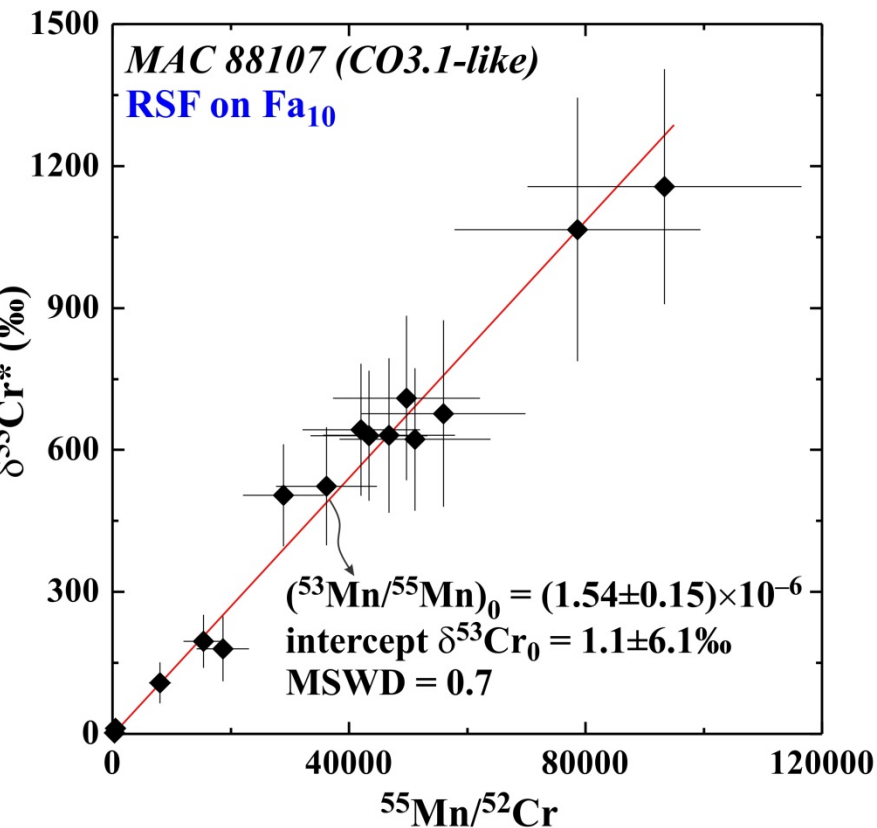


# Mn-Cr relative sensitivity factors (RSF) in ferromagnesian olivine



- San Carlos olivine is not a proper standard for  ${}^{53}\text{Mn}$ - ${}^{53}\text{Cr}$  systematics of fayalite

# $^{53}\text{Mn}$ - $^{53}\text{Cr}$ dating of fayalite formation: RSF effect



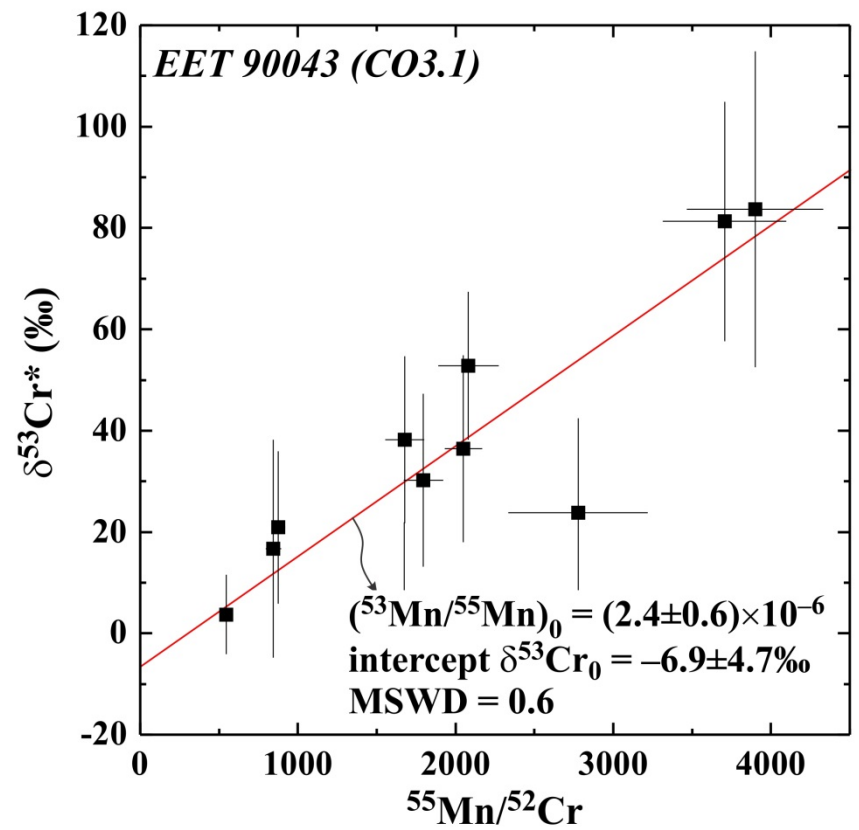
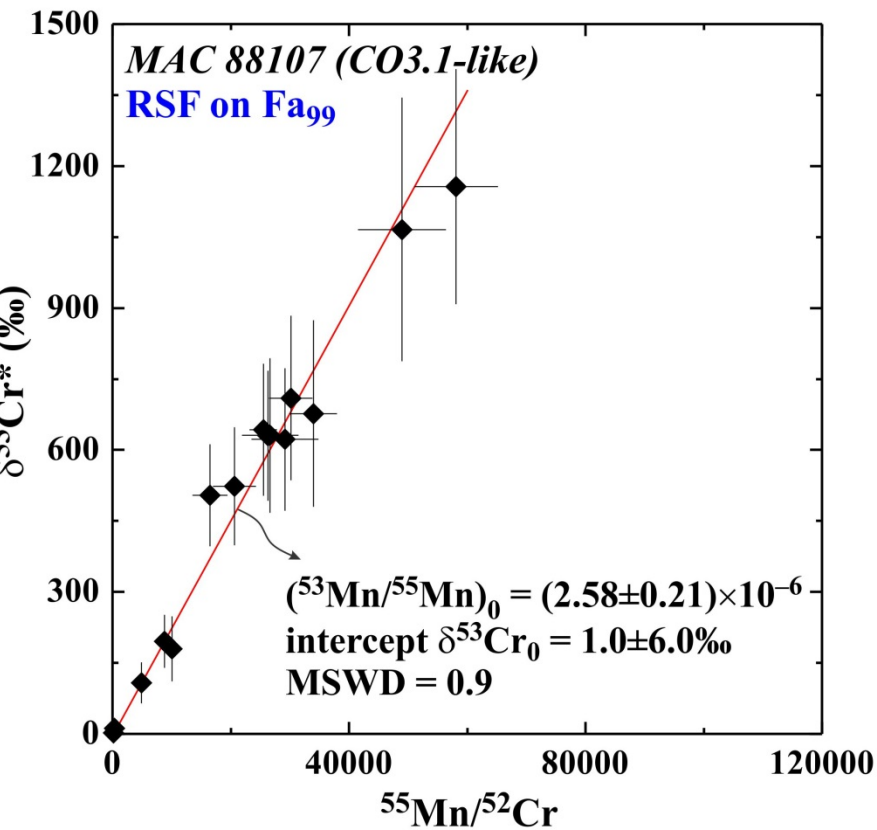
Doyle et al. (2013)

- $7.9^{+0.6}_{-0.5}$  Myr after CV CAIs

- $5.1^{+0.5}_{-0.4}$  Myr after CV CAIs

- San Carlos olivine is not a proper standard for Mn-Cr isotope measurements of fayalite →  $^{53}\text{Mn}$ - $^{53}\text{Cr}$  ages of fayalite published prior to 2013 are incorrect (Hutcheon et al. 1998; Krot et al. 2000; Hua et al. 2005; Jogo et al. 2009 etc.)

# $^{53}\text{Mn}$ - $^{53}\text{Cr}$ dating of fayalite formation in CO & CO-like chondrites



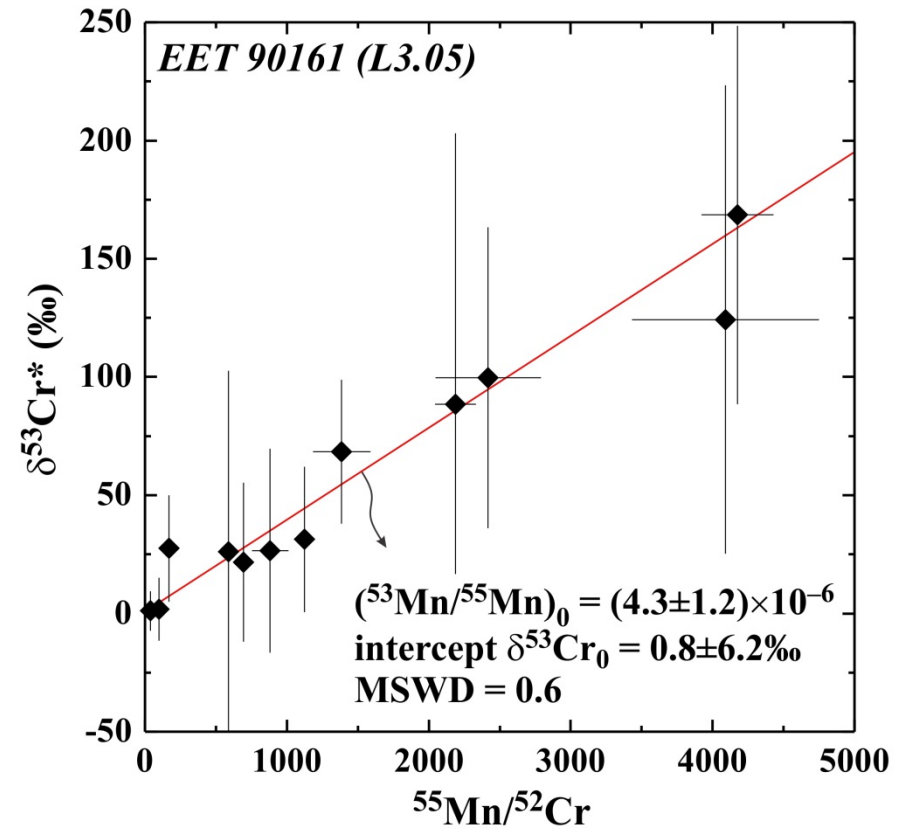
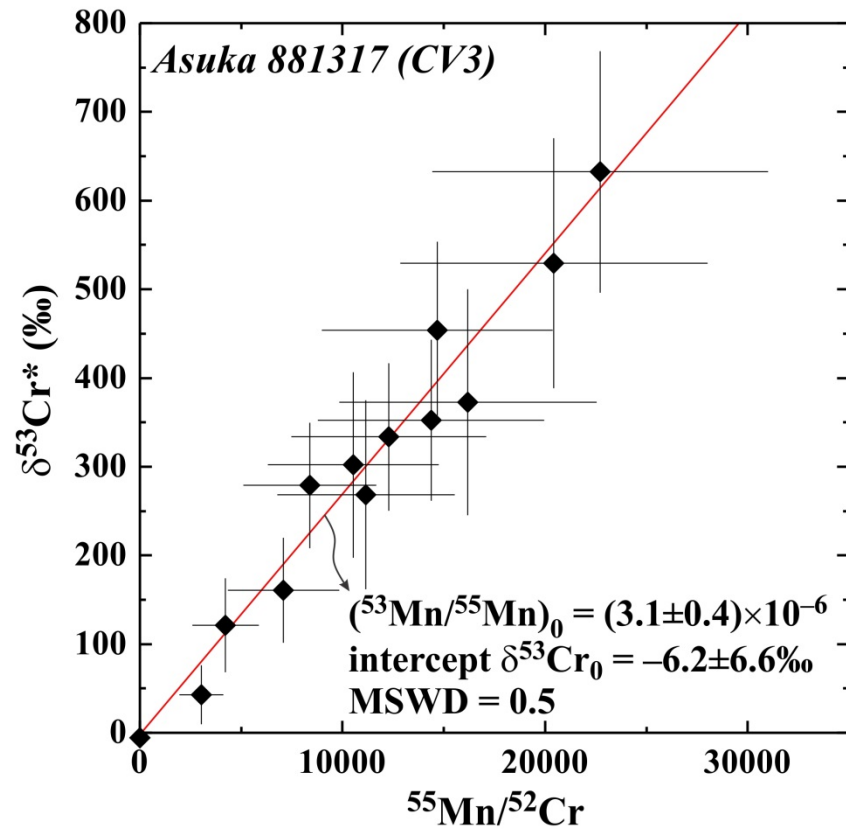
*Doyle et al. (2013)*

•  $5.1^{+0.5}_{-0.4}$  Myr after CV CAIs

•  $5.5^{+1.5}_{-1.2}$  Myr after CV CAIs



# $^{53}\text{Mn}$ - $^{53}\text{Cr}$ dating of fayalite formation in CV & ordinary chondrites



Doyle et al. (2013)

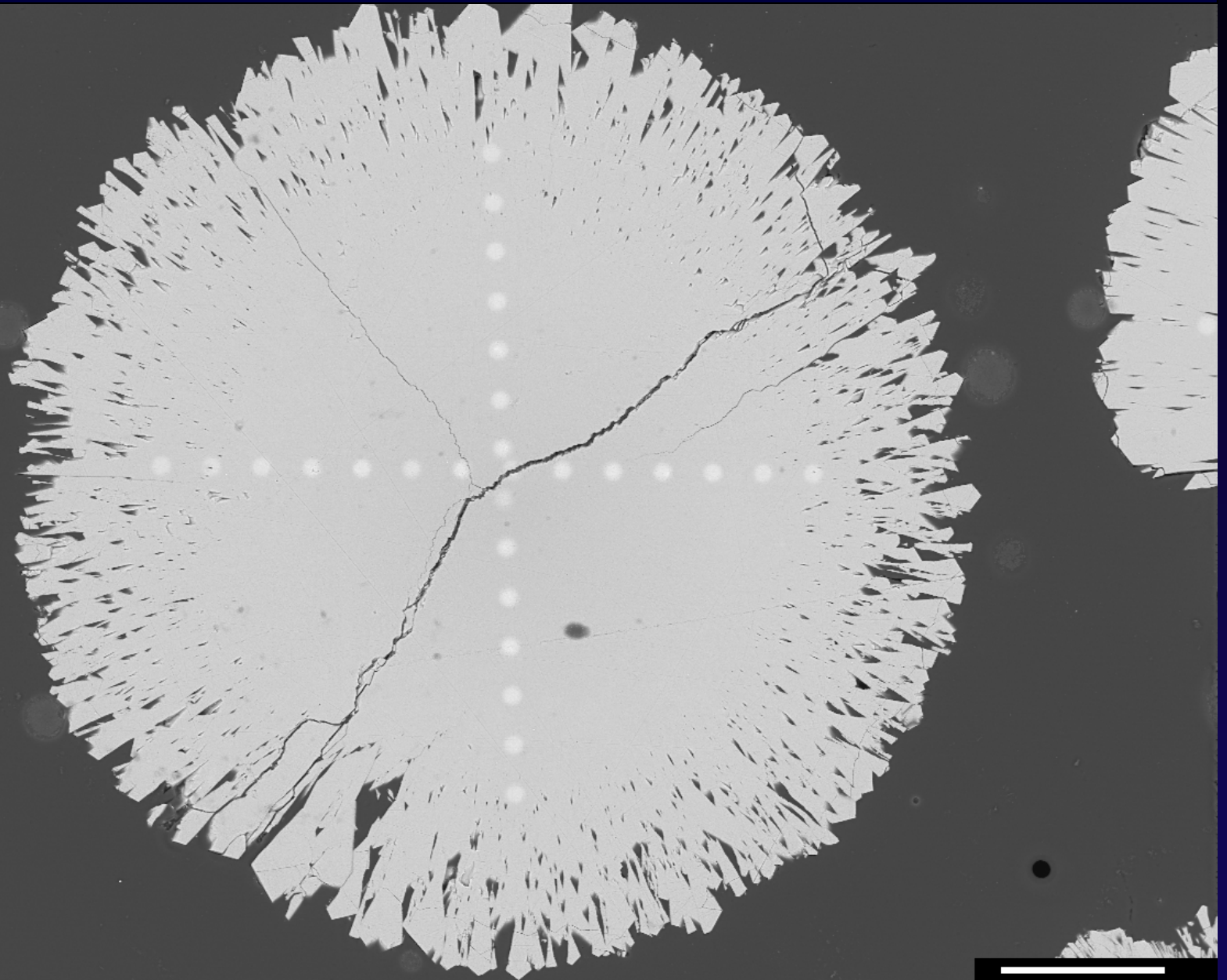
•  $4.2_{-0.7}^{+0.8}$  Myr after CV CAIs

•  $2.4_{-1.3}^{+1.8}$  Myr after CV CAIs





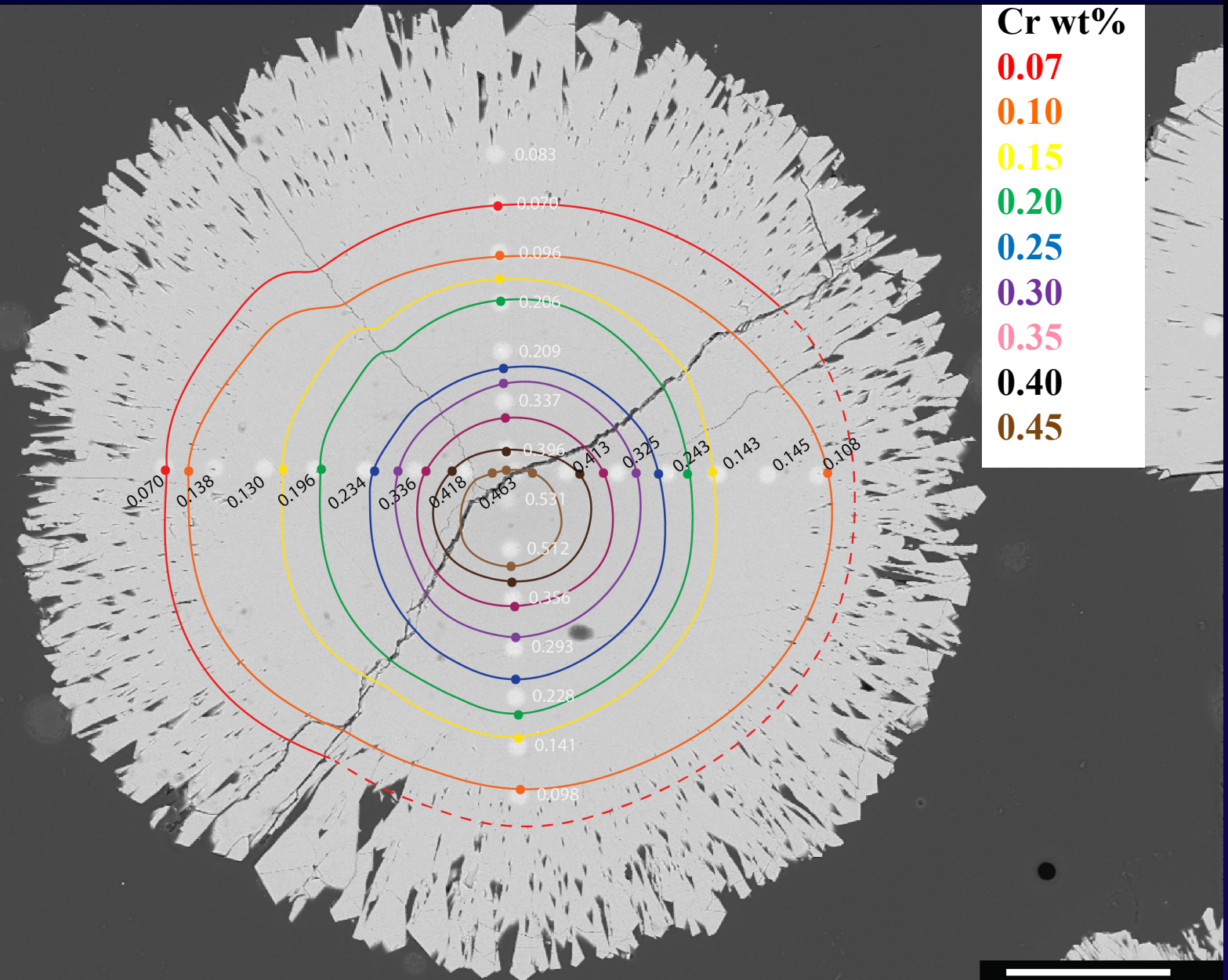
# Synthetic Mn- & Cr-bearing calcite (*N. Sugiura, U. Tokyo*)



*Courtesy of C. Jilly*

100  $\mu\text{m}$

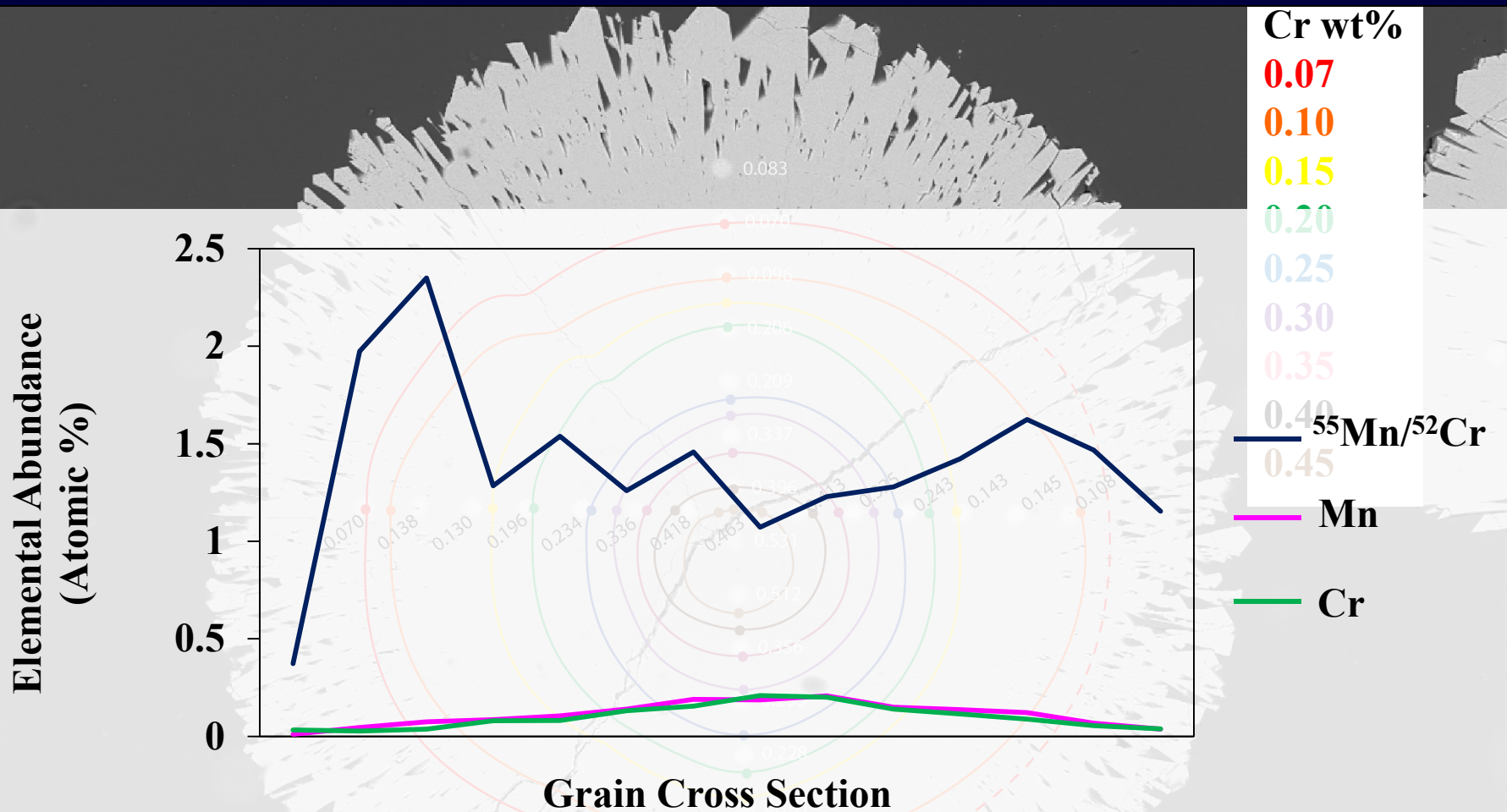
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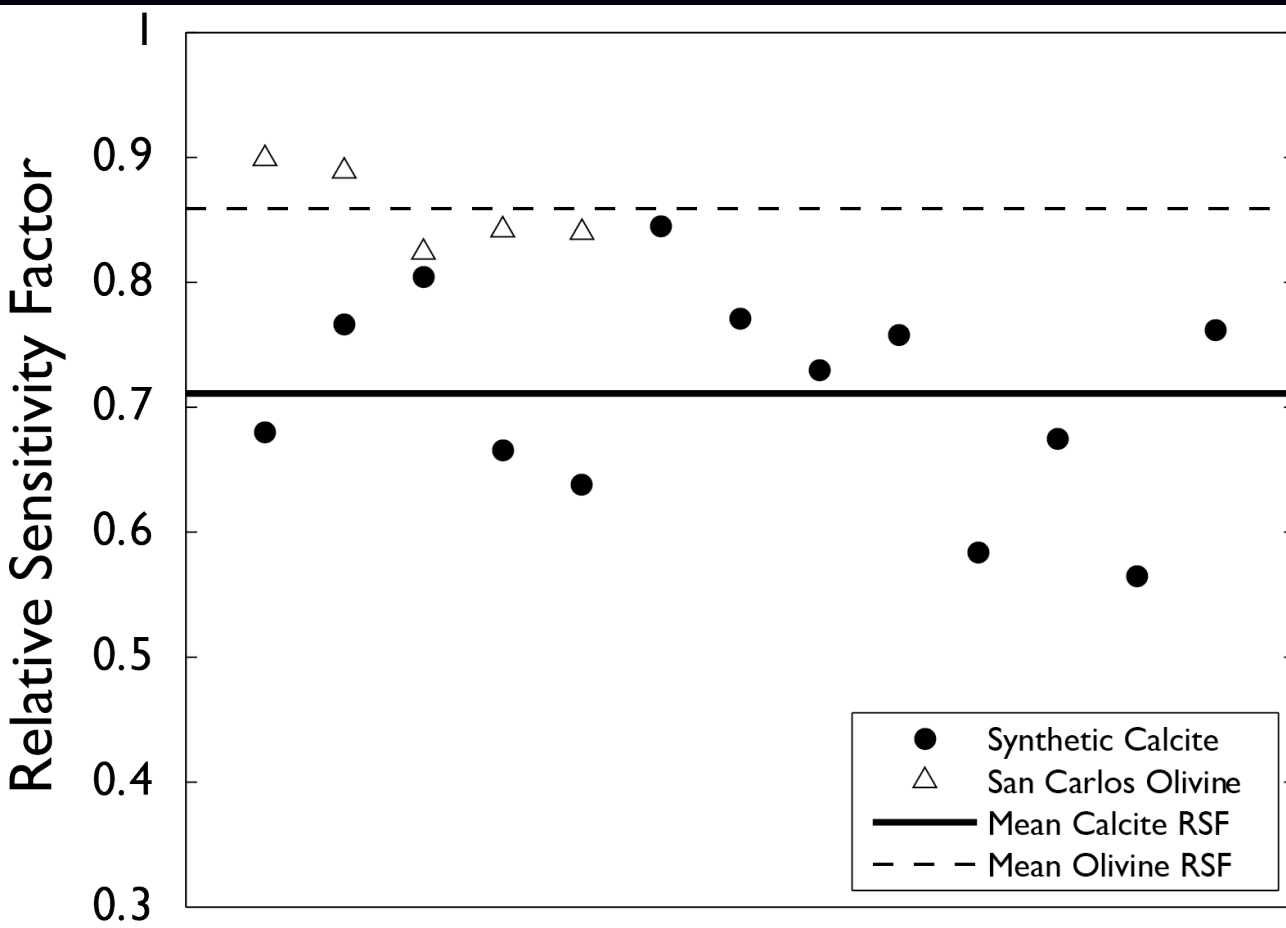
Courtesy of C. Jilly

100  $\mu\text{m}$

# Synthetic Mn- & Cr-bearing calcite (*N. Sugiura, U. Tokyo*)



# Mn-Cr relative sensitivity factors in SC olivine vs. synthetic calcite



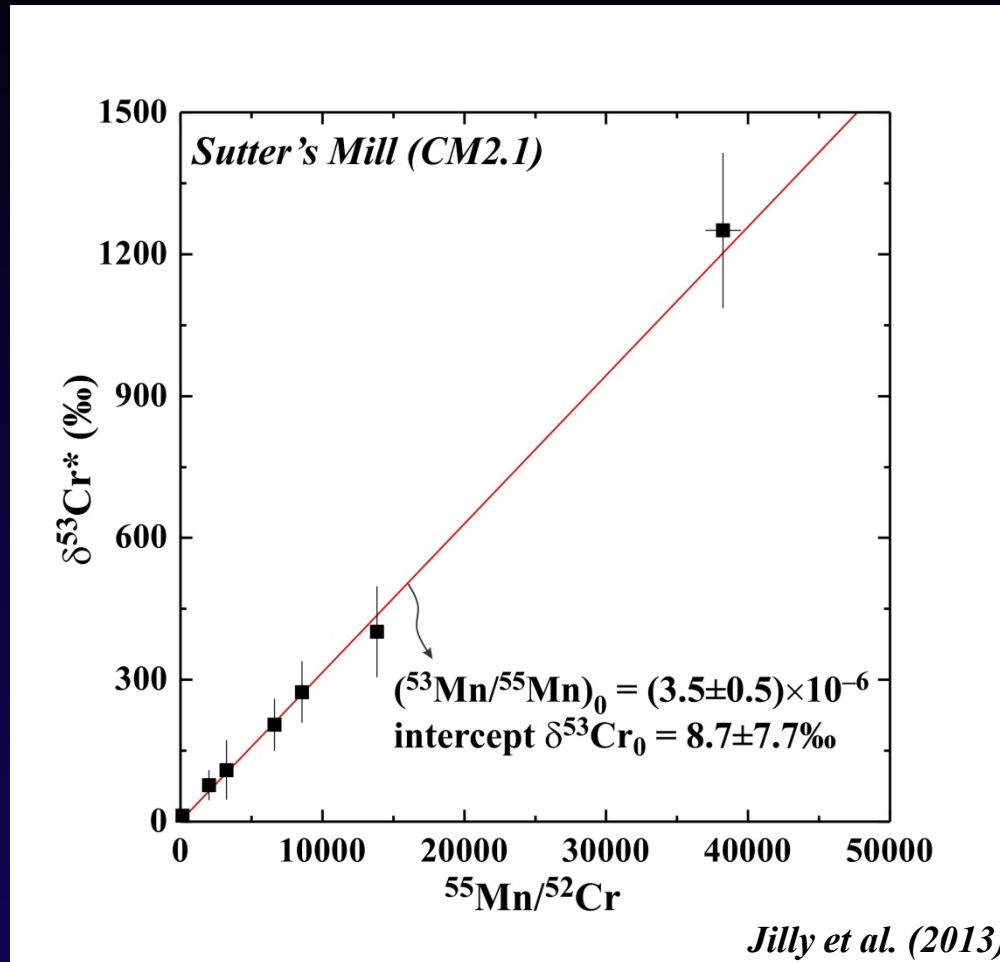
RSF SC olivine ~ 0.86

RSF synth calcite ~ 0.71

*Courtesy of C. Jilly*

- $^{53}\text{Mn}$ - $^{53}\text{Cr}$  dating of carbonates by SIMS requires proper standards
- $^{53}\text{Mn}$ - $^{53}\text{Cr}$  ages of carbonates published prior to 2012 are incorrect
- $^{53}\text{Mn}$ - $^{53}\text{Cr}$  ages of dolomite [Fujiya et al. (2012, 2013)] were measured using calcite standard & should be considered as preliminary

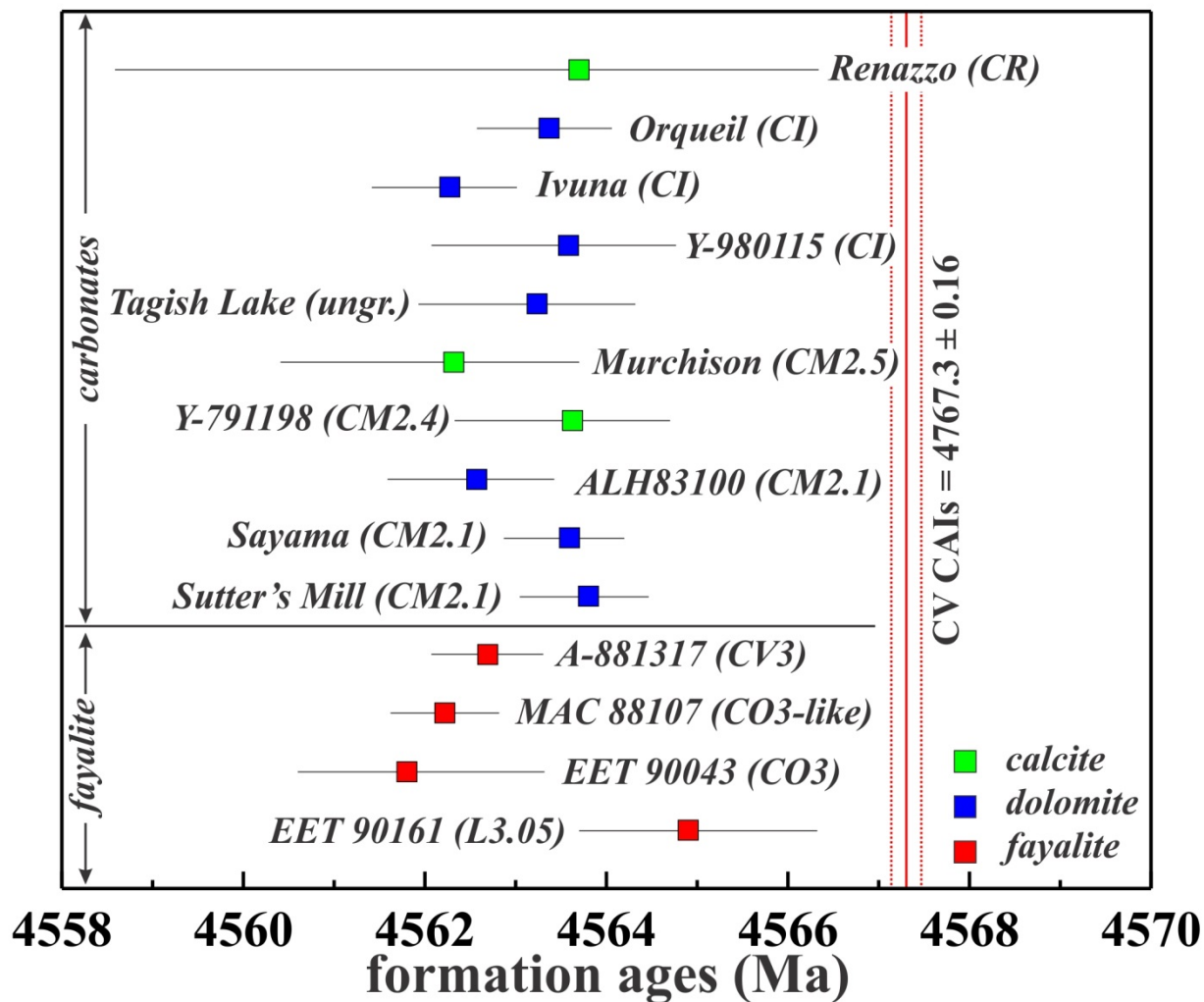
# $^{53}\text{Mn}$ - $^{53}\text{Cr}$ dating of dolomite in CO chondrites



- $3.5_{-0.8}^{+0.7}$  Myr after CV CAIs



# $^{53}\text{Mn}$ - $^{53}\text{Cr}$ ages of fayalite & carbonates in UOCs & CCs



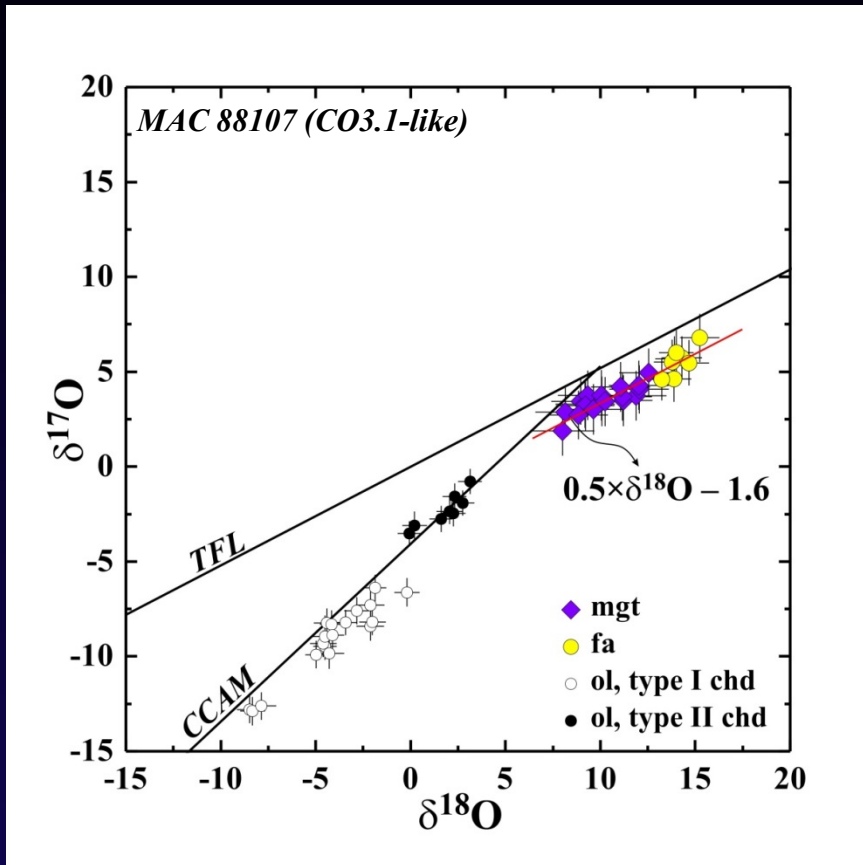
Fujiya et al. (2012, 213); Jilly et al. (2013); Doyle et al. (2013)



Part IV. Oxygen-isotope compositions of aqueously-formed  
minerals & estimated D/H ratios of asteroidal water:  
Constraints on the origin of asteroidal water

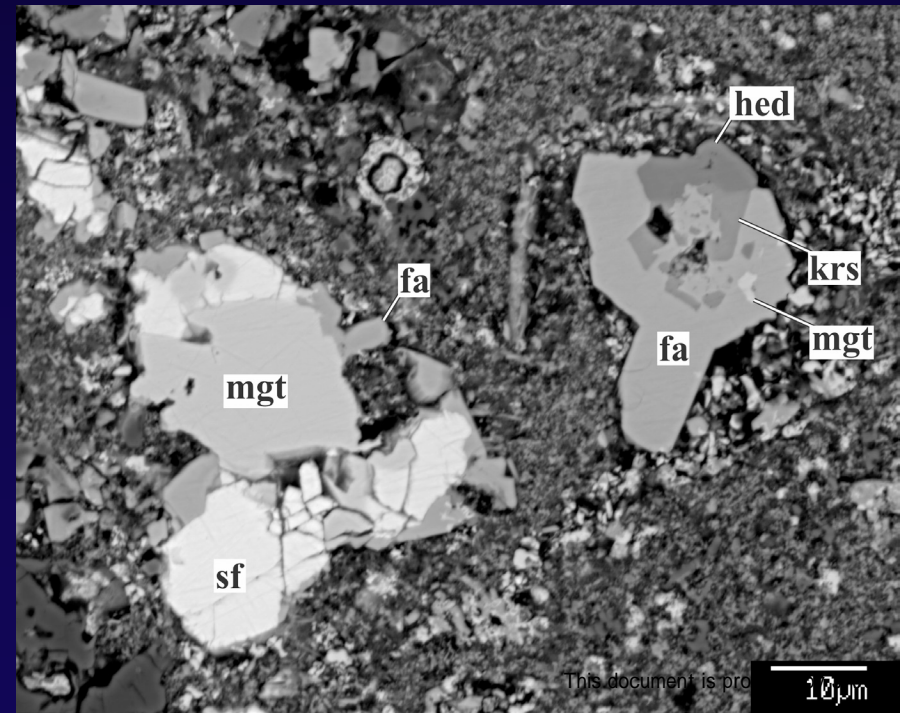


# O-isotope compositions of fayalite & magnetite in MAC 88107



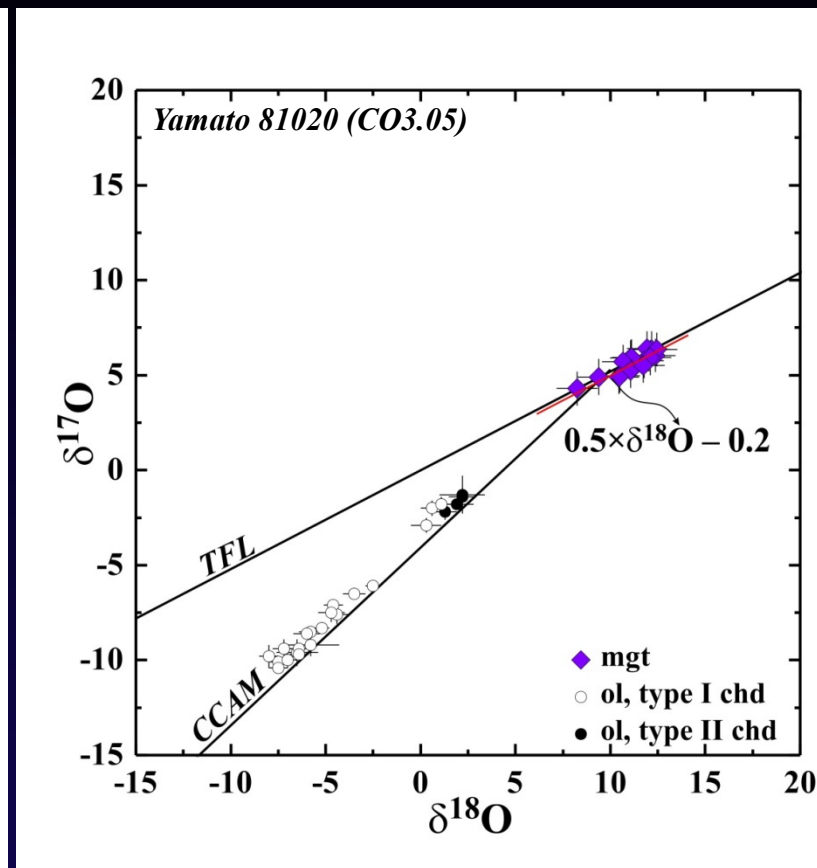
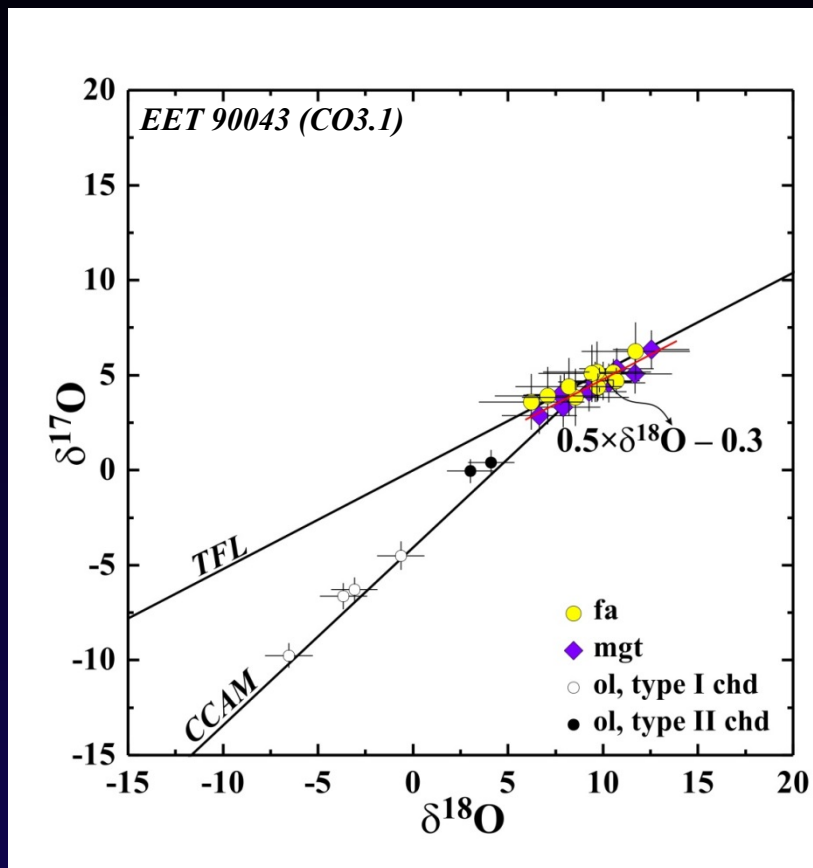
Krot et al. (2013)

- *fa* & *mgt* along slope-0.5 line
  - $\Delta^{17}\text{O} \sim -1.6 \pm 0.9\text{‰}$ , where  $\Delta^{17}\text{O} = \delta^{17}\text{O} - 0.52 \times \delta^{18}\text{O}$
- chondrule *ol*
  - type I:  $\Delta^{17}\text{O} \sim -6.9 \pm 1.8\text{‰}$
  - type II:  $\Delta^{17}\text{O} = -3.2 \pm 0.8\text{‰}$
- *fa* & *mgt* are in oxygen isotopic disequilibrium with chondrule *ol*



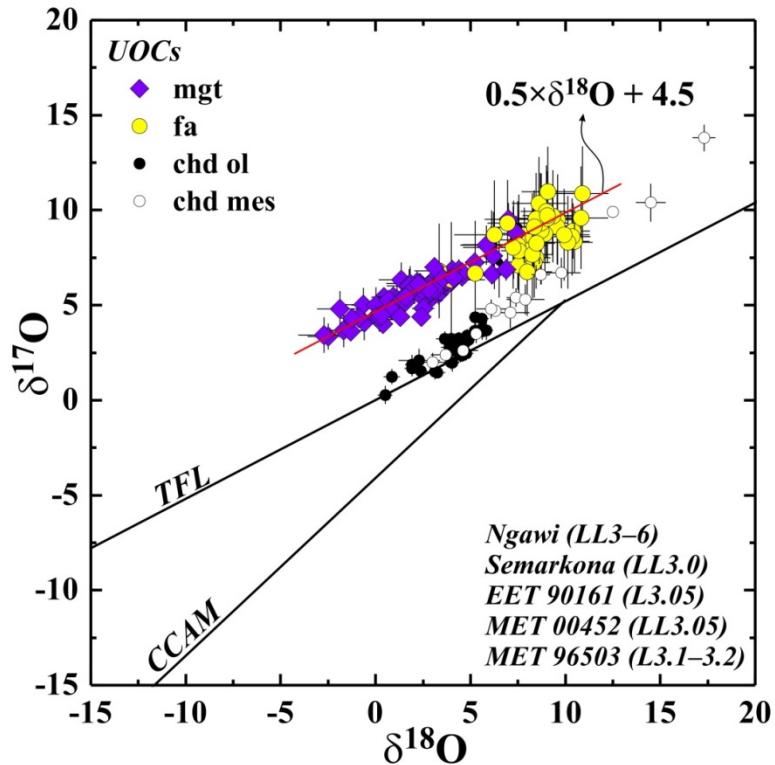


# O-isotope compositions of fayalite & magnetite in CO chondrites



- *fa* & *mgt* along slope-0.5 line
  - $\Delta^{17}\text{O} \sim -0.3 \pm 0.4\text{‰}$
- chondrule *ol*
  - type I:  $\Delta^{17}\text{O} \sim -5\text{‰}$
  - type II:  $\Delta^{17}\text{O} \sim -2\text{‰}$
- *fa* & *mgt* are in oxygen isotopic disequilibrium with chondrule *ol*

# O-isotope compositions of fayalite & magnetite in UOCs

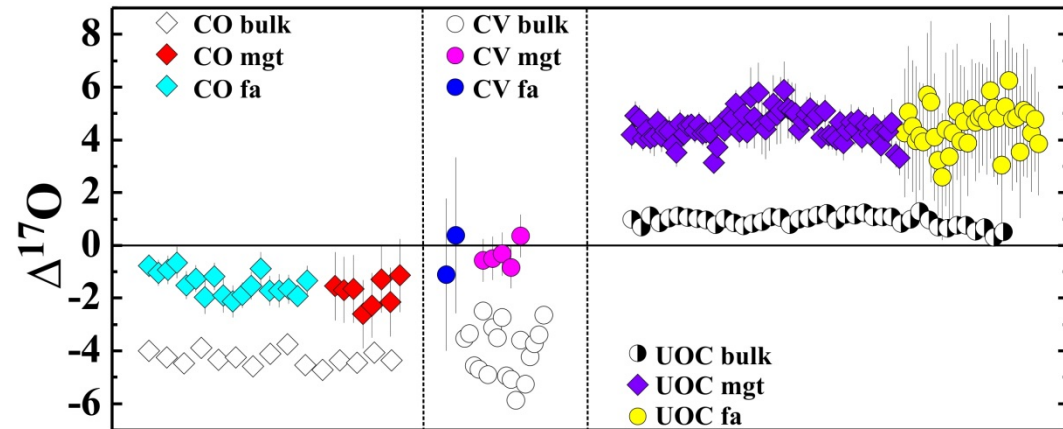
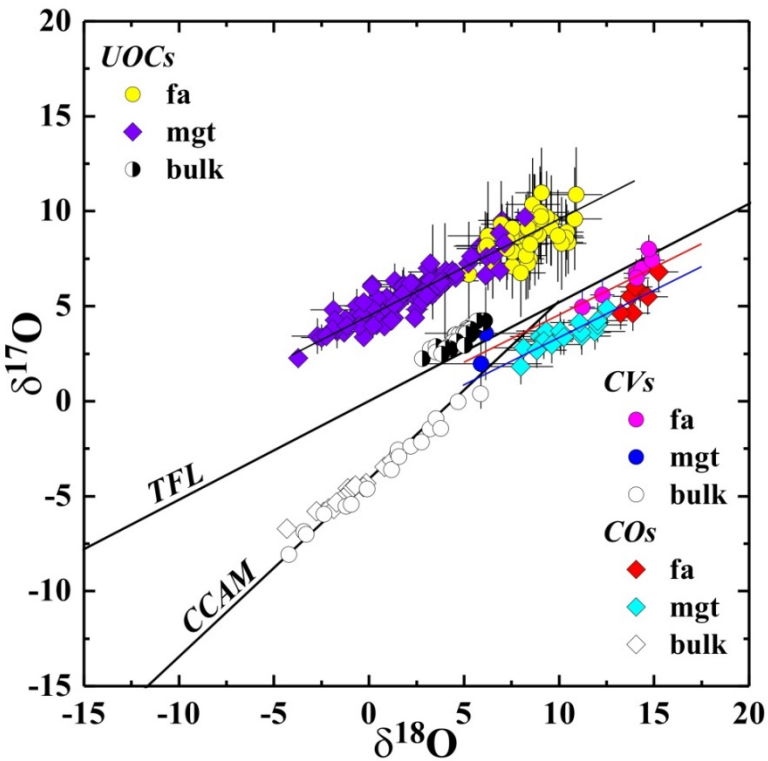


- *fa* & *mgt* along slope-0.5 line
  - $\Delta^{17}\text{O} \sim +4.5\text{‰}$
- chondrule silicates
  - *ol*:  $\Delta^{17}\text{O} \sim +0.6 \pm 0.5\text{‰}$
  - *mes*:  $\Delta^{17}\text{O} +1$  to  $+4.5\text{‰}$
- *fa* & *mgt* in oxygen isotopic disequilibrium with chondrule *ol*
- chondrule *mes* experienced O-isotope exchange with a fluid

Kita et al. (2010); Krot et al. (2012); Doyle et al. (2013)



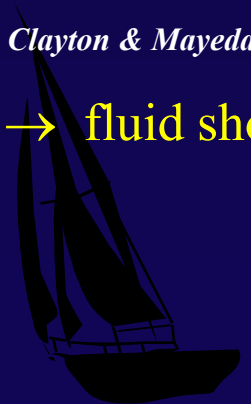
# Oxygen-isotope compositions of aqueous solutions



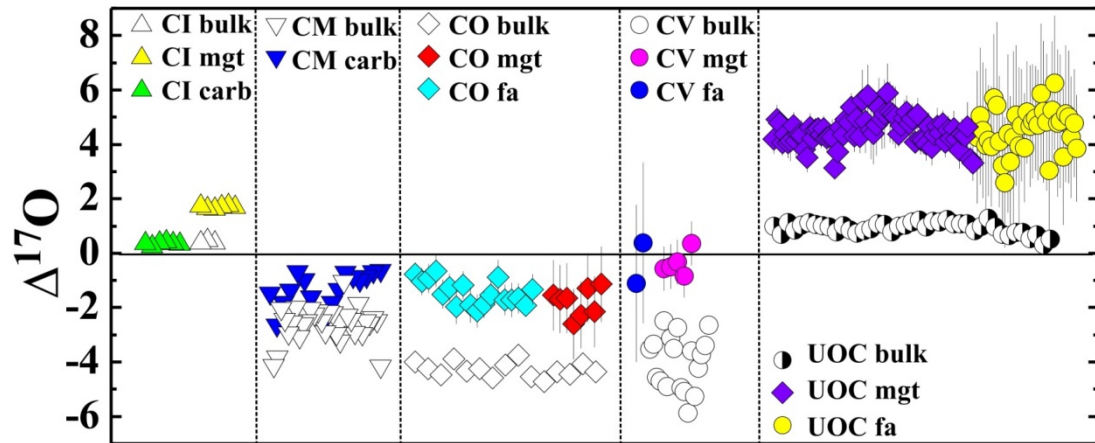
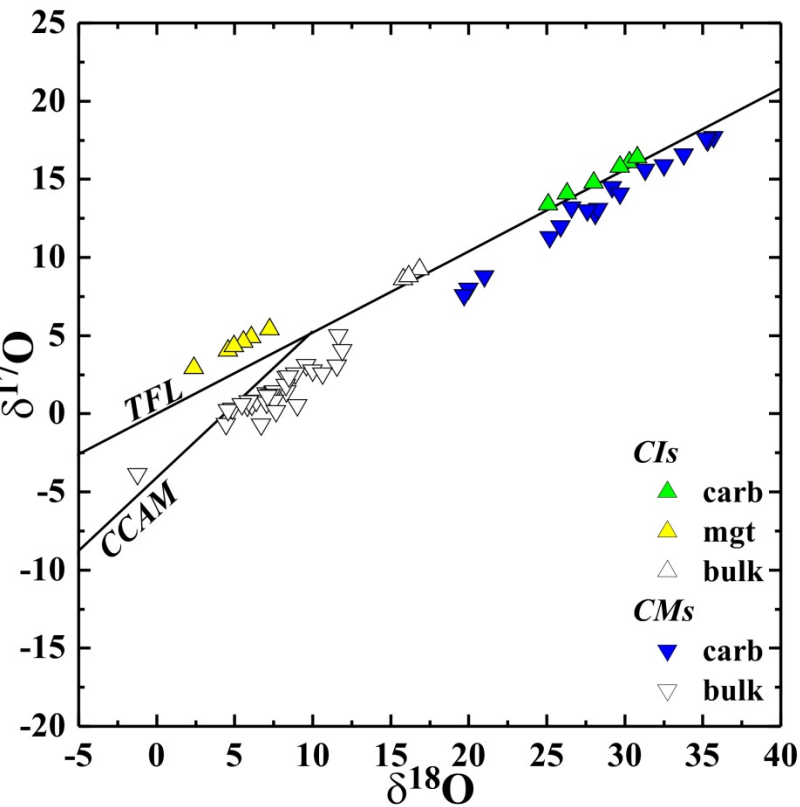
- in UOC, CV & CO:
  - $\Delta^{17}\text{O}$  of *fa* & *mgt* differ from chondrule silicates & bulk chondrites
  - plot along slope-0.5 line
  - $\Delta^{17}\text{O}_{fa \& mgt} = \Delta^{17}\text{O}_{fluid}$

Clayton & Mayeda (1999); Krot et al. (2013); Doyle et al. (2013)

→ fluid shows no evidence for exchange with  $^{16}\text{O}$ -enriched anhydrous silicates



# Oxygen-isotope compositions of aqueous solutions & water ices



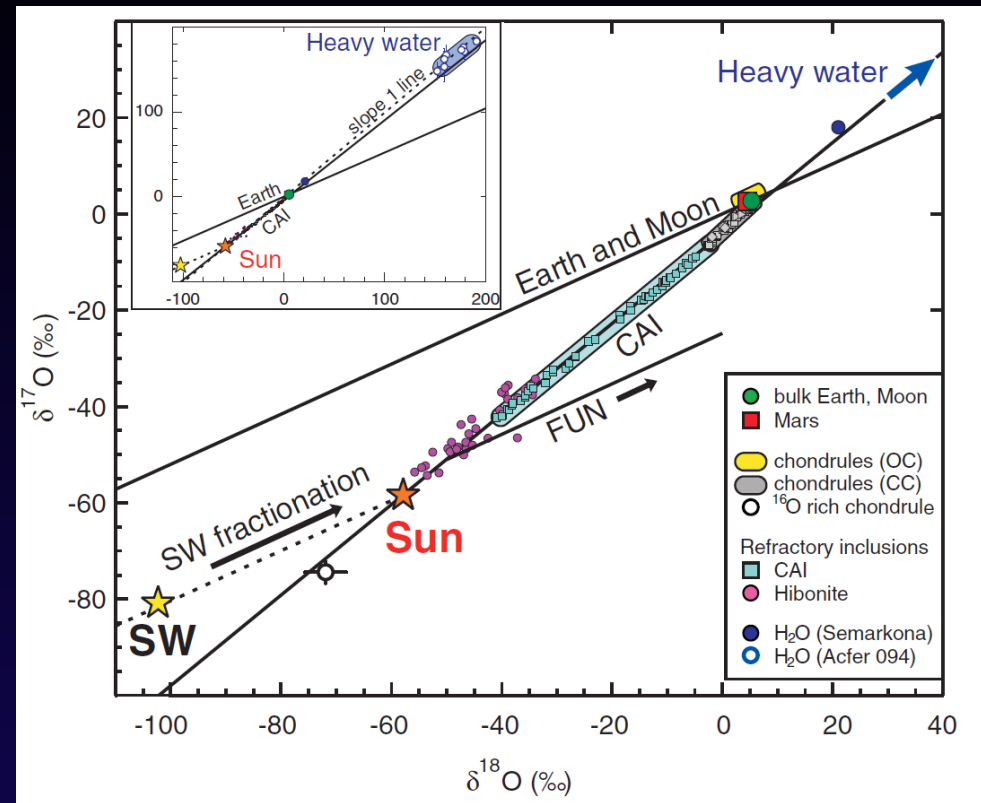
- in CM & CI chondrites:
  - $\Delta^{17}\text{O}$  of *carb* approach  $\Delta^{17}\text{O}$  of whole-rock meteorites
  - small changes in  $\Delta^{17}\text{O}$  value of a fluid with increasing degree of aqueous alteration

Clayton & Mayeda (1999), Rowe et al. (1994), Benedix et al. (2003)  
Leshin et al. (1997, 2011)

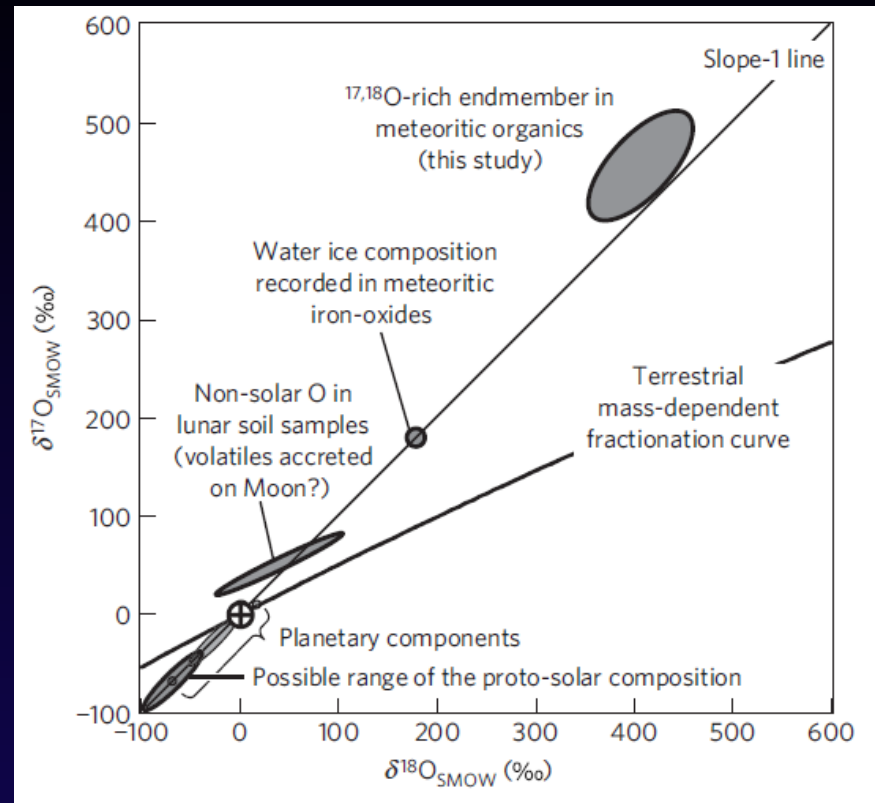
$\Delta^{17}\text{O}$  of *fa* & *mgt* in ordinary, CO & CV, & of *mgt* & *carb* in CI & CM chondrites can be used as a proxy\* for  $\Delta^{17}\text{O}$  values of water ices that accreted into their parent bodies

\* $\Delta^{17}\text{O}$  of water prior to formation of *mgt*, *fa*, & *carb* is not known

# Sources of water ices on chondrite asteroids



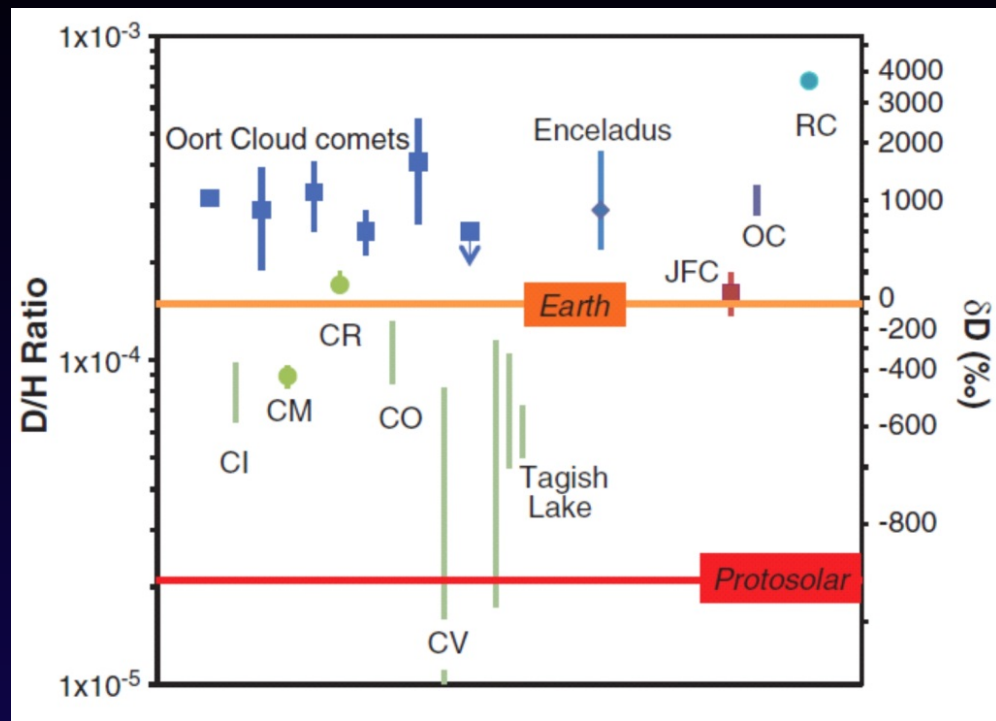
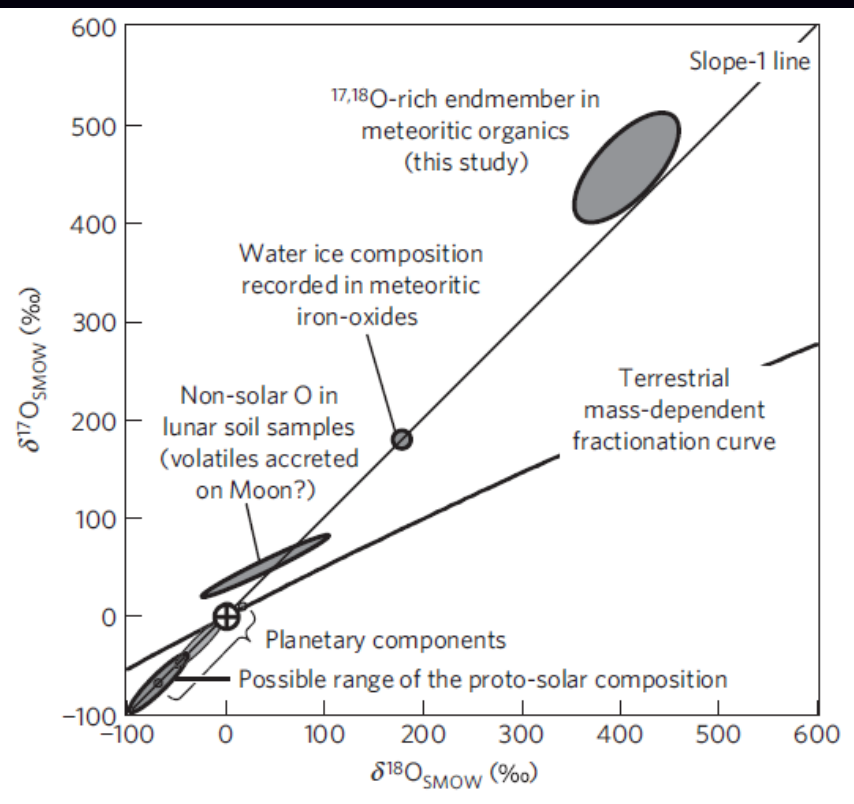
McKeegan et al. (2011)



Hashizume et al. (2011)

- in the CO self-shielding models (Yurimoto & Kuramoto, 2004; Lyons & Young, 2005), water ices in the outer disk is highly  $^{17}\text{O}$  &  $^{18}\text{O}$ -enriched relative to gas & solids in inner disk
  - iron oxides in Acfer 094 (ungrouped):  $\Delta^{17}\text{O} \sim +90\text{‰}$  (Sakamoto et al., 2007)
  - grains in IOM from Y-793495 (CR):  $\Delta^{17}\text{O} \sim +500\text{‰}$  (Hashizume et al., 2007)
- water ices on OC & CC asteroids were not anomalously  $^{17}\text{O}$ - &  $^{18}\text{O}$ -enriched, suggesting a local origin

# Sources of water ices on chondrite asteroids



Alexander et al. (2012)

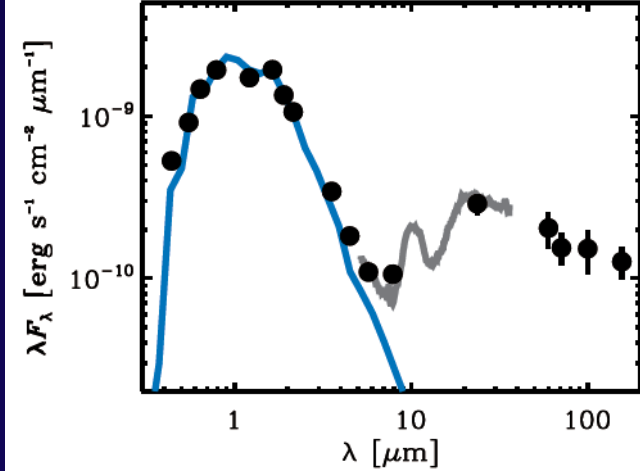
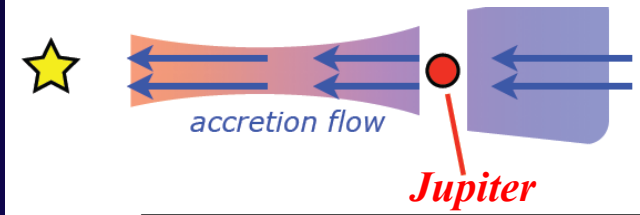
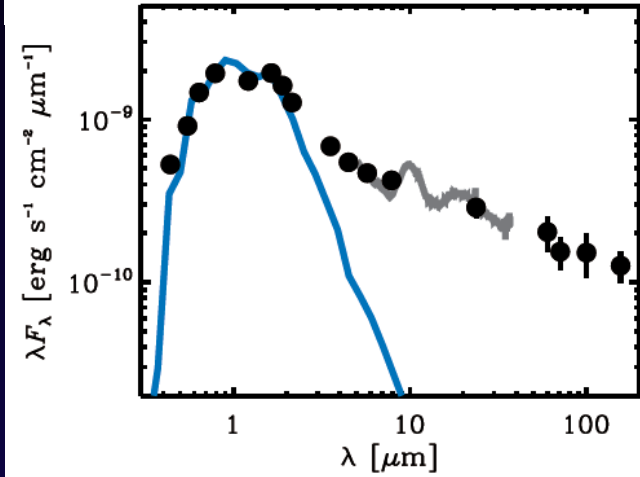
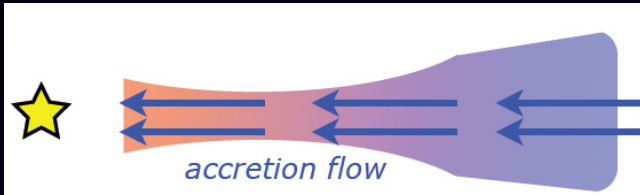
Hashizume et al. (2011)

- local, inner Solar System origin of asteroidal water ices is consistent with the inferred D/H ratios of chondritic water (Alexander et al. 2012), which are very different from D/H ratios in Oort Cloud Comets measured



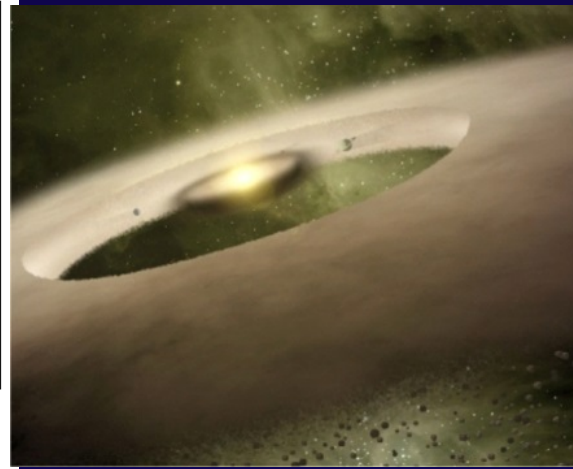
# Role of Jupiter in isolating inner disk from the outer Solar System

*full, optically thick disk*



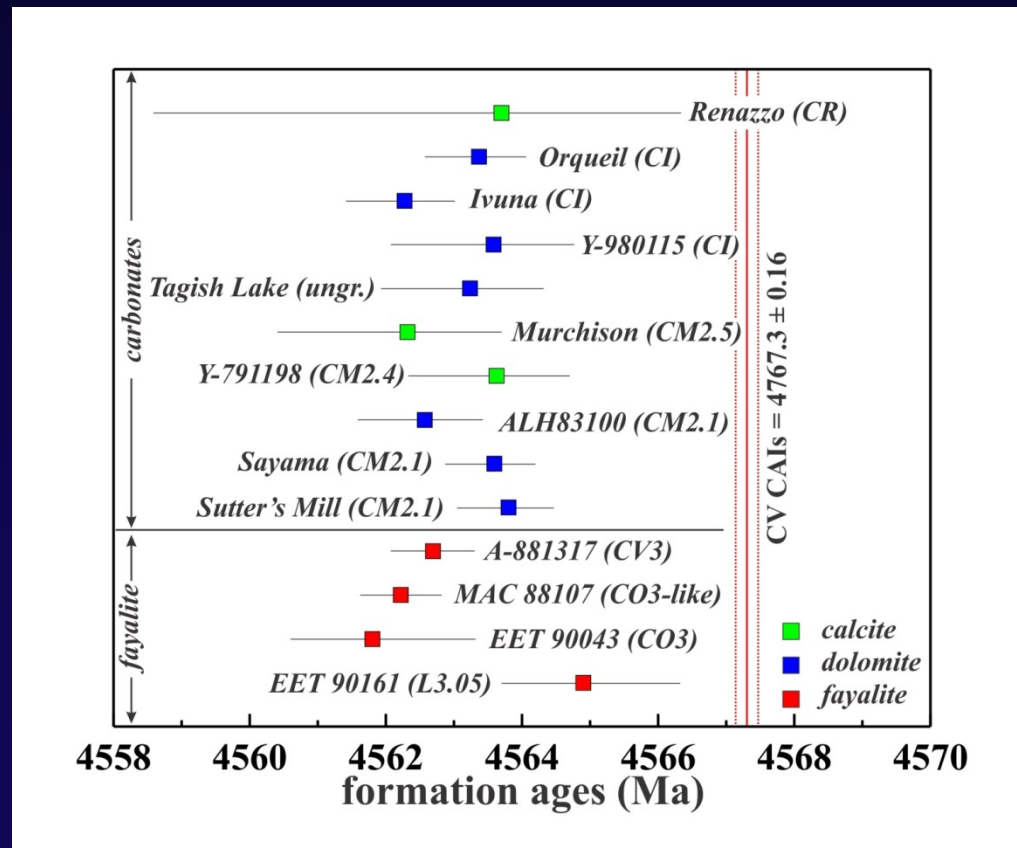
- low influx of water from the outer Solar System  $\geq 2.5$  Myr after CAI formation could be due to an early growth of Jupiter that opened a gap in the disk & prevented significant radial transport of dust from outside its orbit

*pre-transitional disk with a gap*



# Conclusions (I)

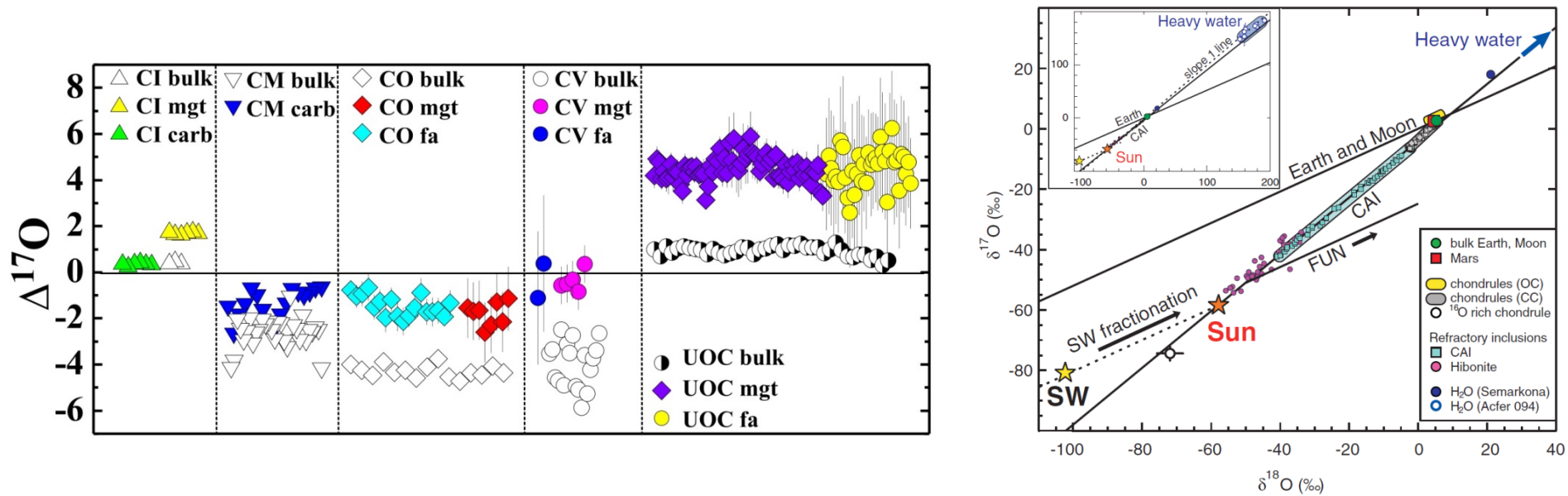
- CI, CM, CR, CV, CO, & UOCs experienced various degrees of aqueous alteration
- $^{53}\text{Mn}$ - $^{53}\text{Cr}$  dating of carbonates in CI, CM, & CR chondrites, & fayalite in CV, CO, & UOCs indicate aqueous activity occurred  $\sim 3$ – $5$  Myr after CV CAIs
- old ages of aqueous alteration & their narrow range are consistent with heating of chondrite asteroids by  $^{26}\text{Al}$ , & inconsistent with a significant role of impact heating





# Conclusions (II)

- near the terrestrial O-isotope compositions of fayalite, magnetite, & carbonates in UOCs & CCs imply that asteroidal water had preferentially a local, inner Solar System origin (*Alexander et al. 2012*)



- ordinary chondrites accreted isotopically heavier water than CCs & may have had a higher proportion of the outer Solar System water
- low influx of water from the outer SS during accretion of CC asteroids could be due to an early growth of Jupiter that opened a gap in the disk & prevented significant radial transport of dust from outside its orbit

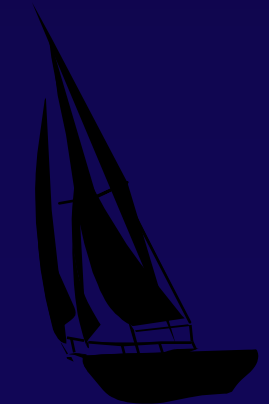
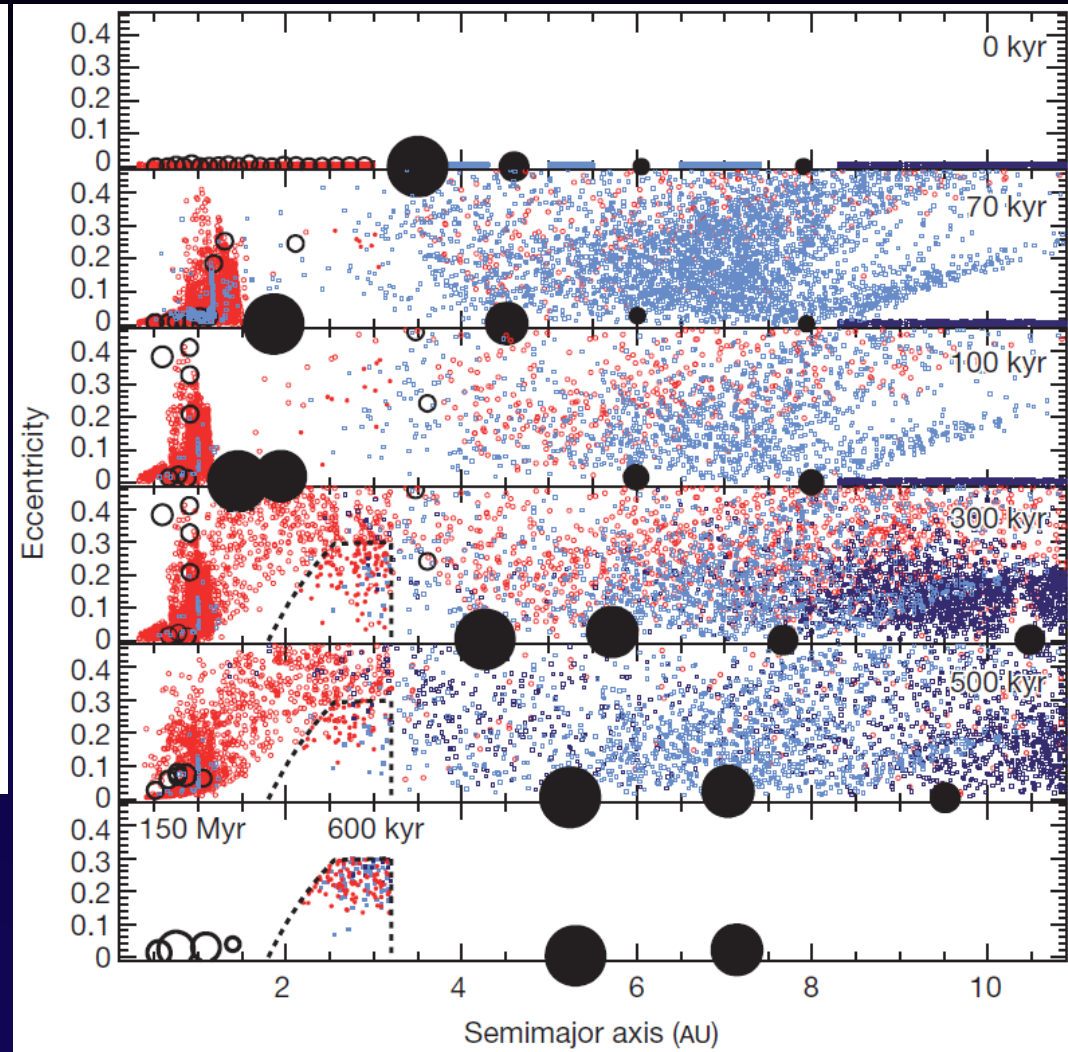
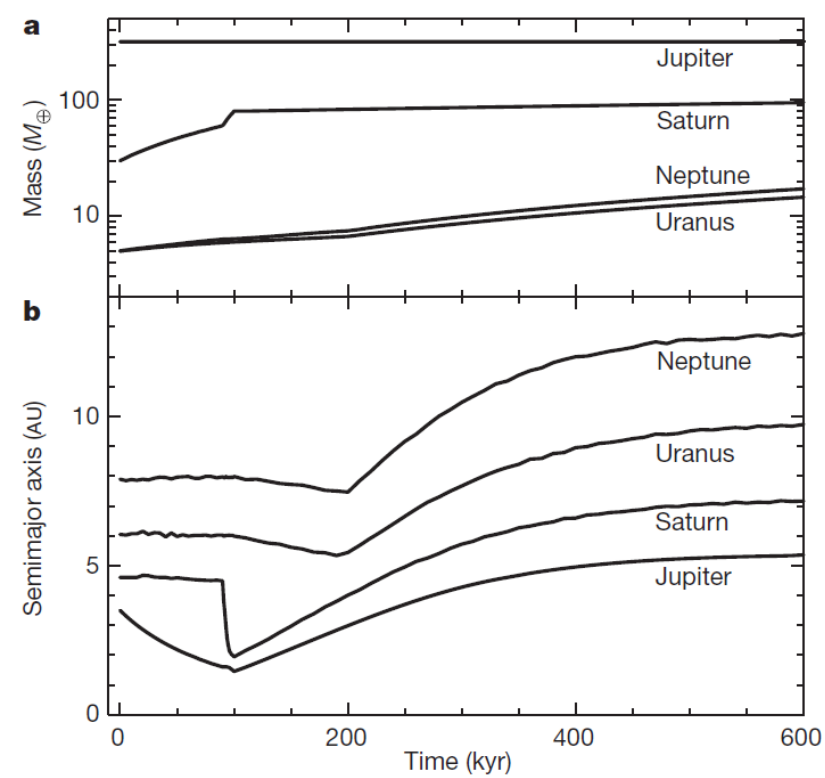
# Extra slides



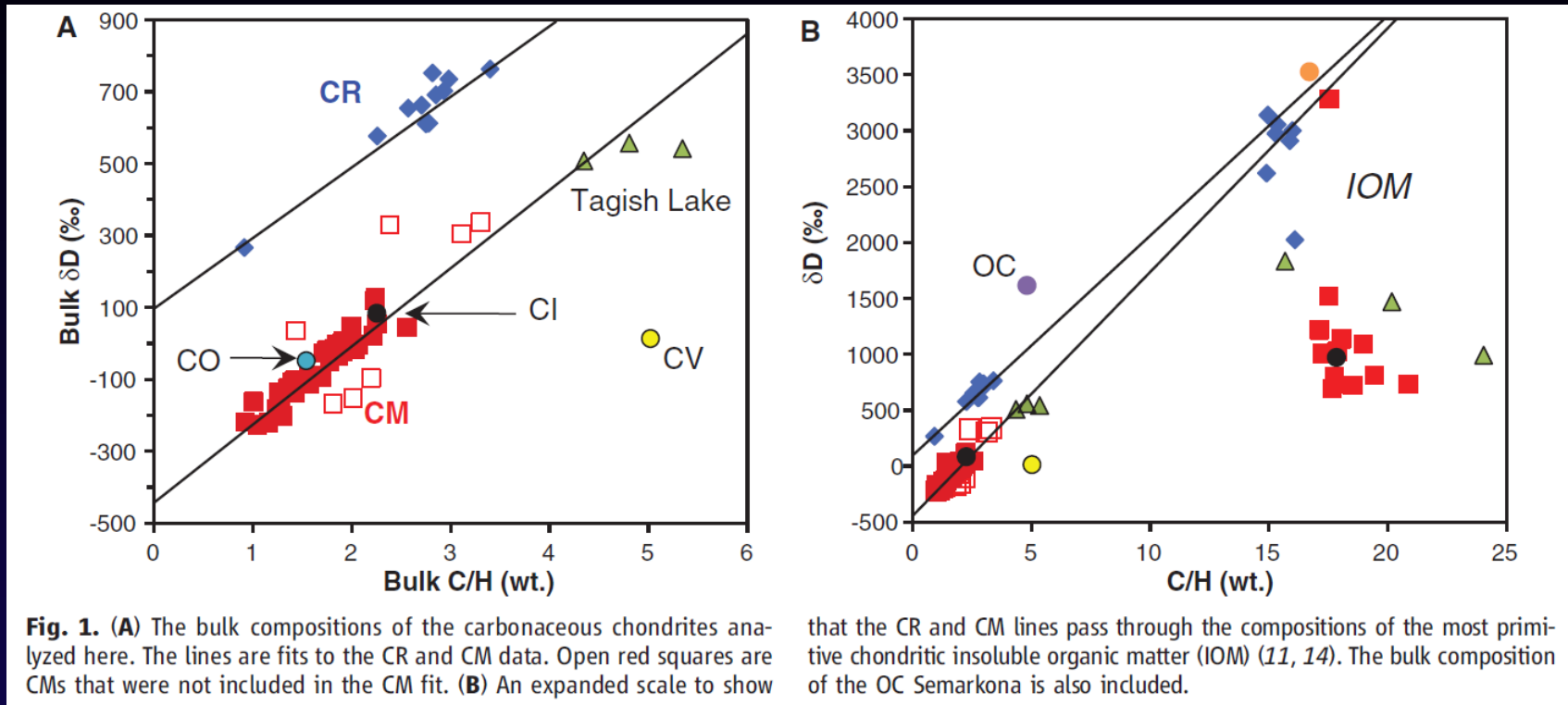
# A low mass for Mars from Jupiter's early gas-driven migration

Kevin J. Walsh<sup>1,2</sup>, Alessandro Morbidelli<sup>1</sup>, Sean N. Raymond<sup>3,4</sup>, David P. O'Brien<sup>5</sup> & Avi M. Mandell<sup>6</sup>

*Nature (2011)*



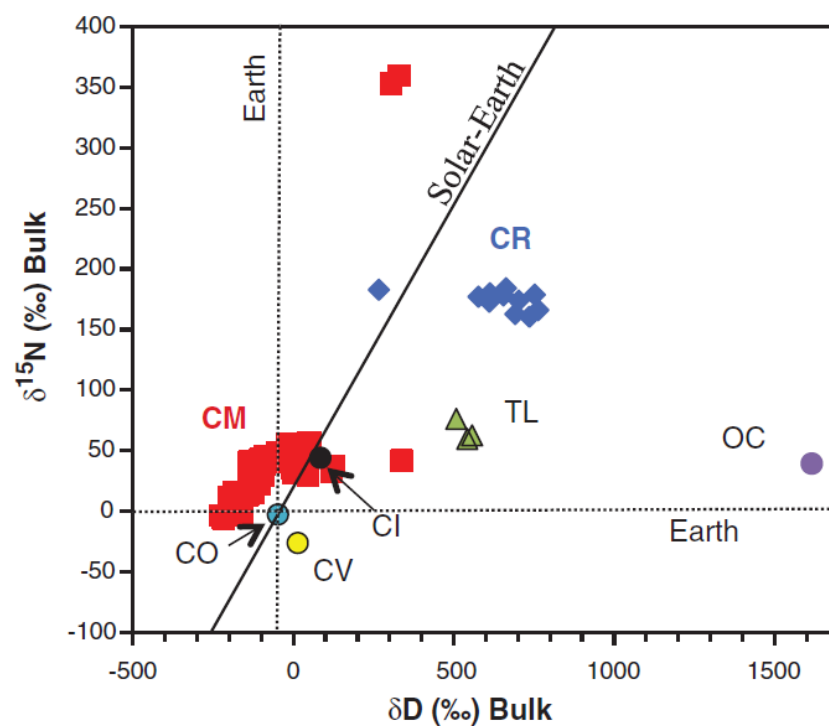
# Sources of water on asteroids: Insights from H & N isotopes



- CM & CR chondrites experienced different degrees of aqueous alteration
- bulk H-isotopic compositions of a chondrite = hydrated silicates + organics
- $\delta D$  vs. C/H  $\rightarrow$  isotopic composition of water;  $\delta D = (D/H_{\text{sample}}/D/H_{\text{SMOW}} - 1) \times 1000$
- CM & CR trends converge on the region where the most primitive insoluble organic matter (IOM) plots  $\rightarrow$  share a common primitive organic component



**Fig. 3.** The bulk hydrogen and nitrogen isotopic compositions of chondrites (TL, Tagish Lake). The line connects the solar and terrestrial isotopic compositions. Bodies similar to chondrites are potential sources of Earth's volatiles. For reasons discussed in the text, CI-like material with ~10% contributions of material with isotopically solar compositions, but a roughly chondritic H/N value, can most simply explain Earth's bulk hydrogen and nitrogen isotopic compositions (3).



- if comets were not the sources of Earth's water, asteroids become the most likely candidates, along with some nebular gas
- most of H & N (as well as other volatiles) accreted in CI-like material along with ~10% contribution to both elements from material with a solar isotopically solar compositions

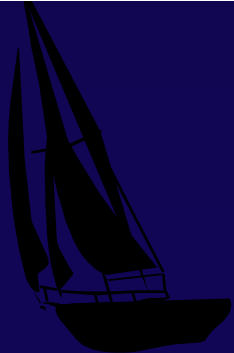


# The Provenances of Asteroids, and Their Contributions to the Volatile Inventories of the Terrestrial Planets

C. M. O'D. Alexander,<sup>1\*</sup> R. Bowden,<sup>2</sup> M. L. Fogel,<sup>2</sup> K. T. Howard,<sup>3,4</sup> C. D. K. Herd,<sup>5</sup> L. R. Nittler<sup>1</sup>

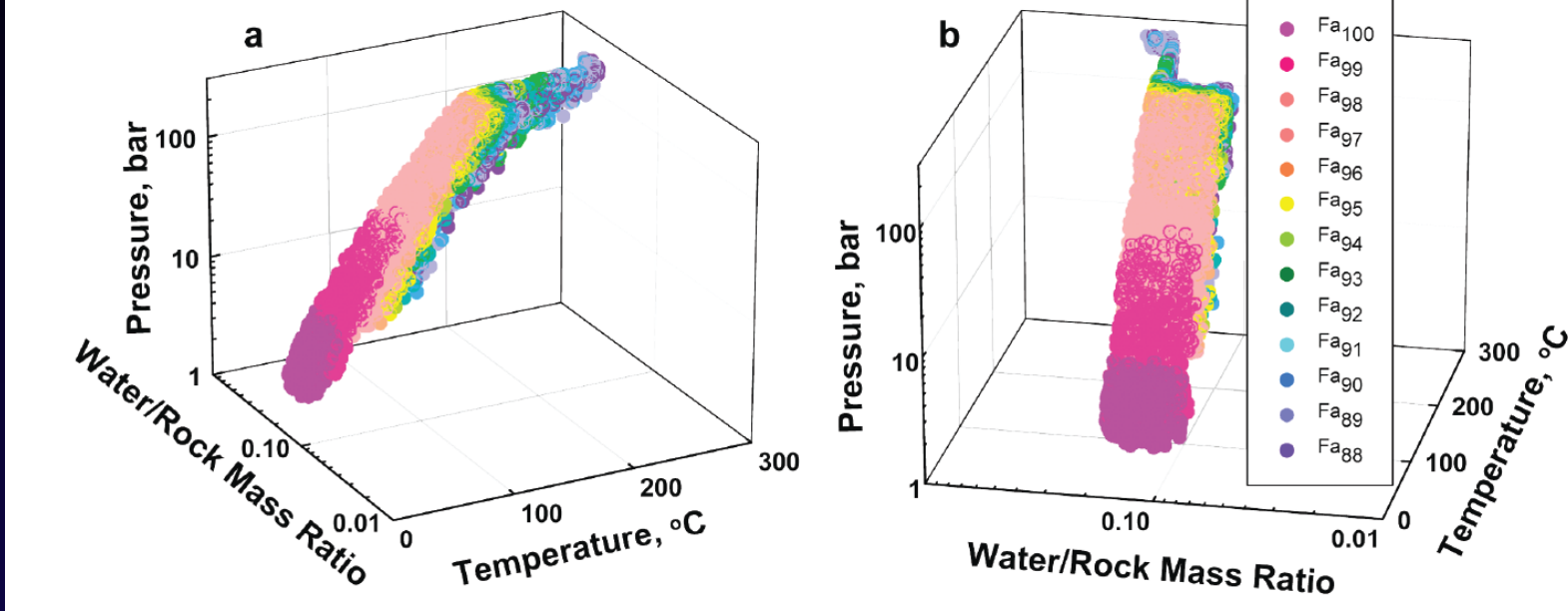
*Science, 337, 721 (2012)*

Determining the source(s) of hydrogen, carbon, and nitrogen accreted by Earth is important for understanding the origins of water and life and for constraining dynamical processes that operated during planet formation. Chondritic meteorites are asteroidal fragments that retain records of the first few million years of solar system history. The deuterium/hydrogen (D/H) values of water in carbonaceous chondrites are distinct from those in comets and Saturn's moon Enceladus, implying that they formed in a different region of the solar system, contrary to predictions of recent dynamical models. The D/H values of water in carbonaceous chondrites also argue against an influx of water ice from the outer solar system, which has been invoked to explain the nonsolar oxygen isotopic composition of the inner solar system. The bulk hydrogen and nitrogen isotopic compositions of CI chondrites suggest that they were the principal source of Earth's volatiles.



# Origin of fayalite-bearing assemblages

Zolotov et al. (2006)



- fayalite & magnetite formed *in situ*, during fluid-rock interaction at water/rock ratio  $\sim 0.1\text{--}0.2$  & temperatures  $\sim 100\text{--}300^\circ\text{C}$
- differences in  $\Delta^{17}\text{O}$  of fayalite+magnetite in UOCs & CCs

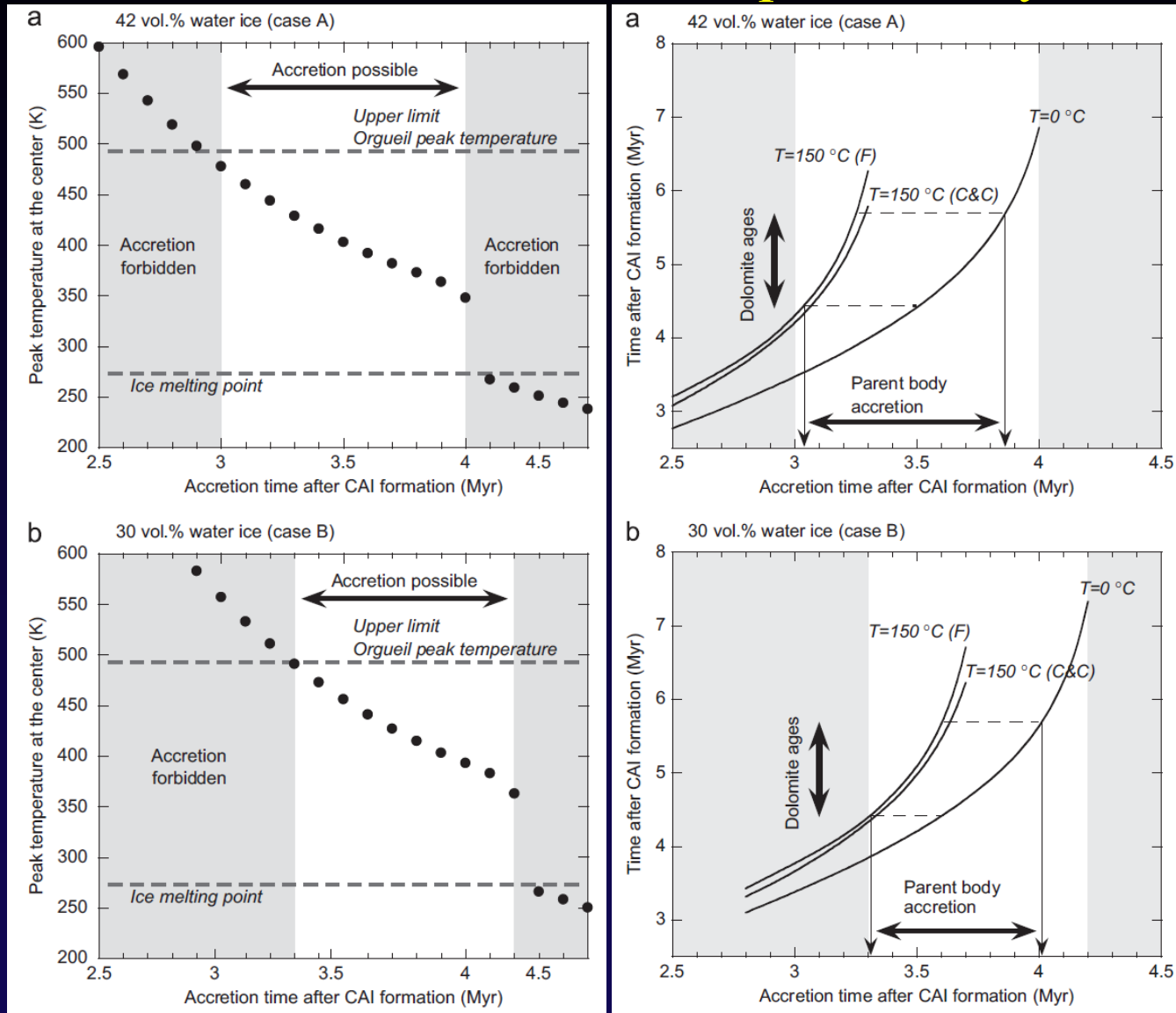
UOC:  $+4.5 \pm 1.1\text{‰}$

CV & CO:  $-0.4 \pm 0.9\text{‰}$

CO-like:  $-1.6 \pm 0.9\text{‰}$

may reflect spatial or temporal variations in O-isotope compositions of water ices that accreted into their parent asteroids

# Accretion time of the CI parent body



Fujiya et al. (2013)

- CIs accreted 3–4 Myr after CV CAIs: based on peak temperature experienced by CIs &  $^{53}\text{Mn}$ - $^{53}\text{Cr}$  ages & formation temperature of carbonates



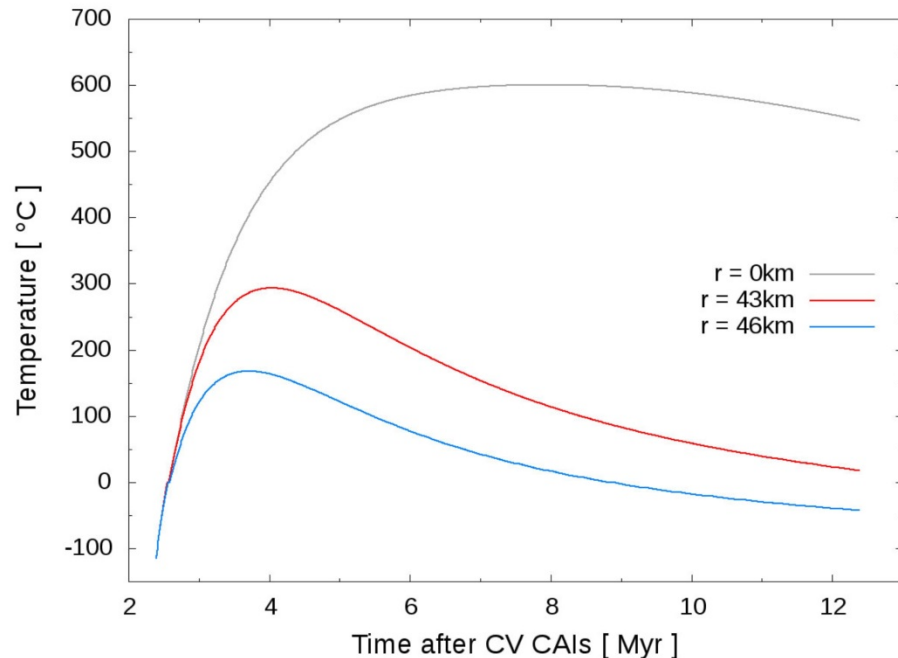
# Accretion ages of CV & CO asteroids

- COs & CVs define metamorphic sequences of petrologic types 3.0–3.7 & 3.1–3.6 with peak metamorphic temperature of  $\sim 600^{\circ}\text{C}$  (*Huss et al. 2006; Bonal et al. 2006, 2007*)
  - $^{26}\text{Al}$  major heating source of asteroids
  - $(^{26}\text{Al}/^{27}\text{Al})_0$  in the disk after epoch of CAI formation could have been uniform at  $\sim 5 \times 10^{-5}$  (*e.g., Schrader et al. 2013*)
- peak metamorphic temperatures can be used to constrain accretion ages of CV & CO parent asteroids

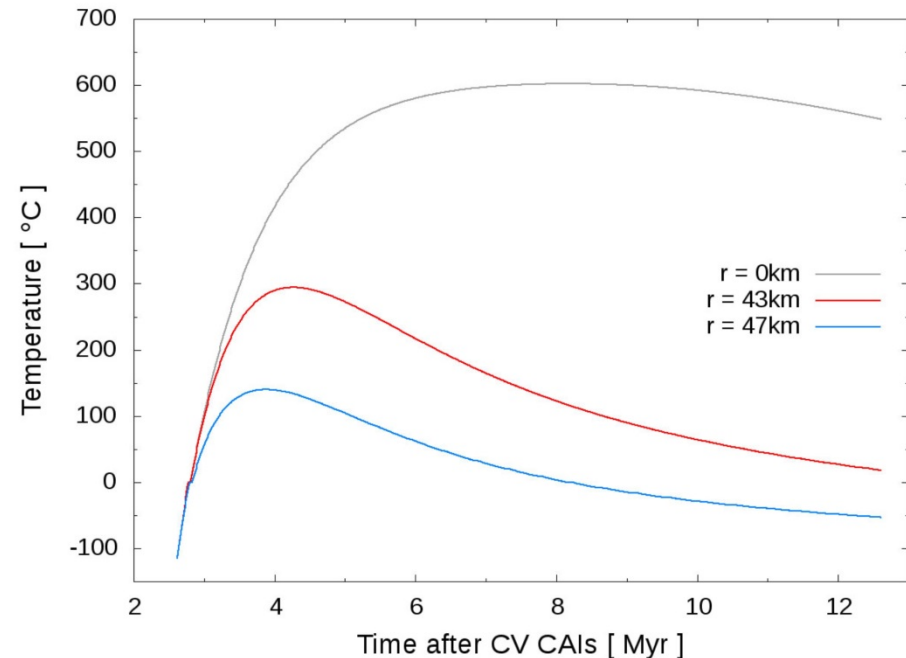


# Accretion ages of CV & CO asteroids & formation of fayalite

CO: 50km, 2.4 Myr after CV CAIs

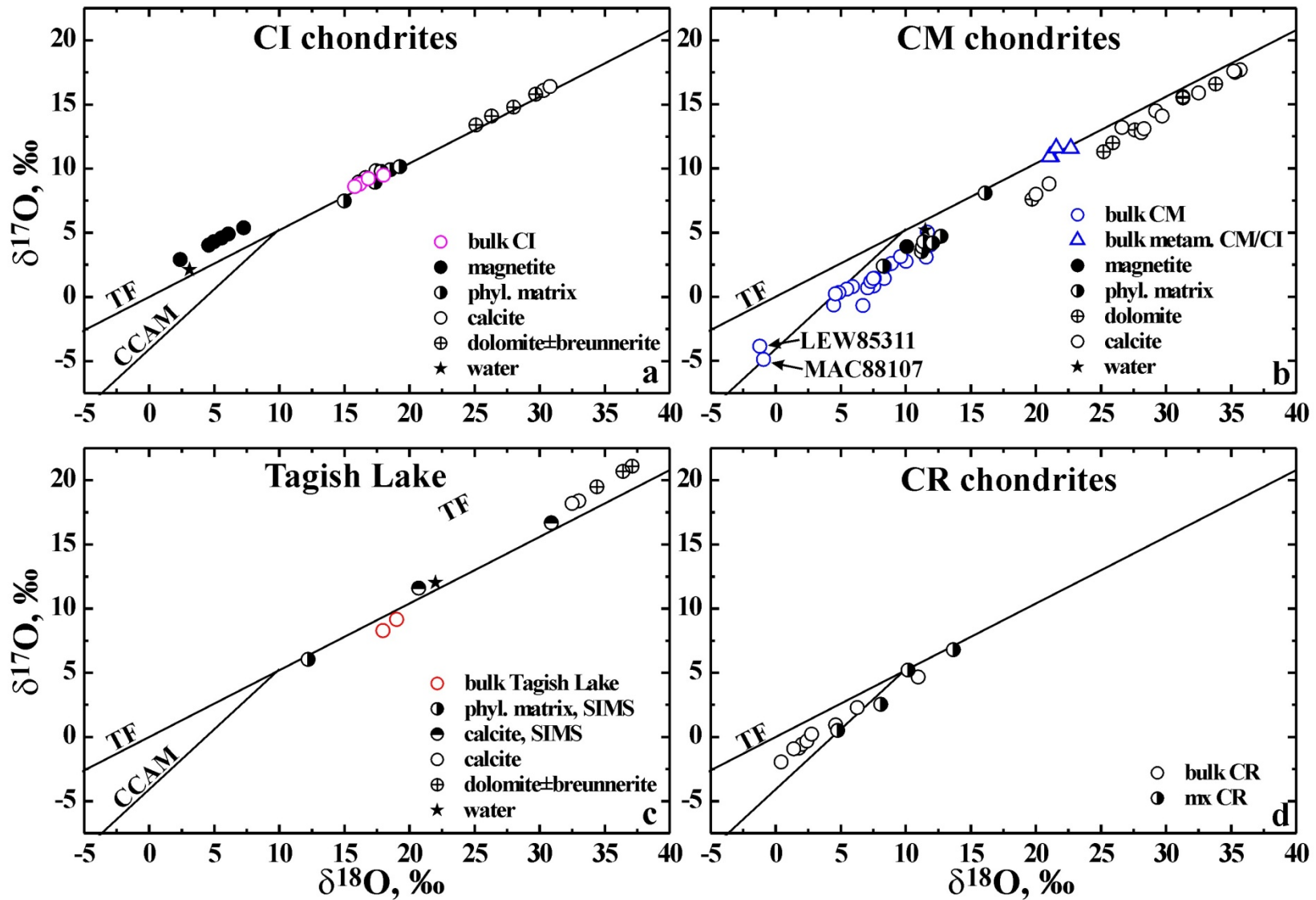


CV: 50km, 2.6 Myr after CV CAIs

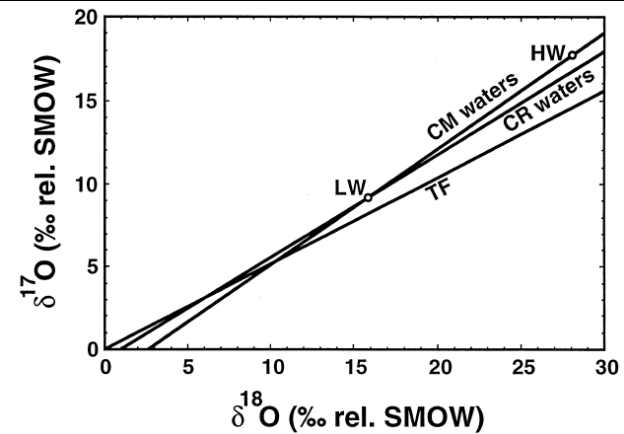


- numerical modeling of thermal history of CV & CO like asteroids with radius of 50 km, water/rock ratio 0.2, & peak metamorphic temperature  $\sim 600^{\circ}\text{C}$
- CV & CO asteroids must have accreted within  $\leq 2.6$  Myr after CV CAIs, consistent with  $^{26}\text{Al}$ - $^{26}\text{Mg}$  ages of CO chondrules ( $2.8 \pm 0.8$  Myr after CAIs, *Kita & Ushikubo 2012*)
- CV & CO fayalite could have precipitated  $< 5$  Myr after CAIs in the outer portions of these asteroids
- CO & CV asteroids accreted with water ices shortly after chondrule formation
- CI & CM parent bodies accreted 3–4 Myr after CAIs (*Fujiya et al. 2012, 2013*)

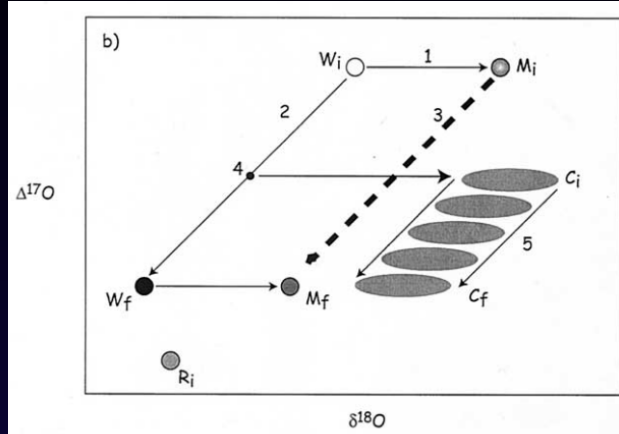
# Chondritic water as a potential record of influx of ice from the outer into the inner Solar System



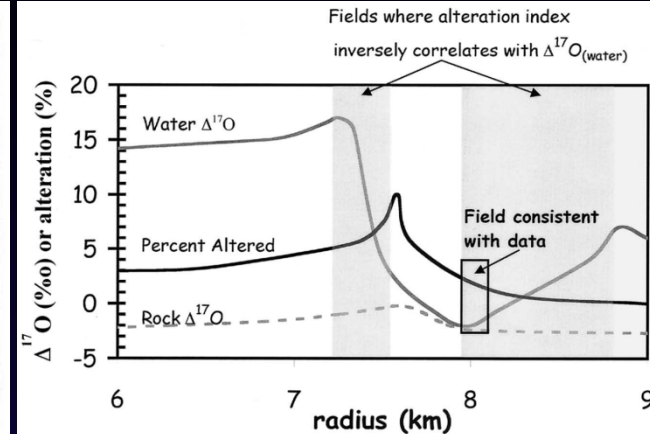
# Oxygen-isotope composition of water ices in CCs



Clayton & Mayeda (1999)



Benedix et al. (2003)



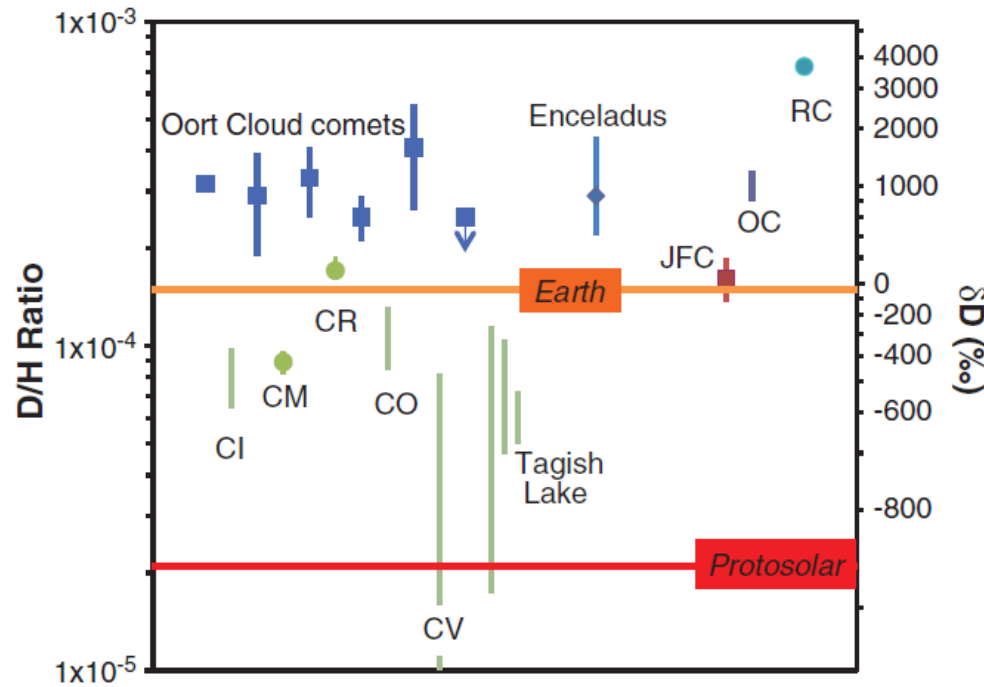
## Previous estimates of the initial $\Delta^{17}\text{O}$ values of water ices in carbonaceous chondrites

- CMs & CRs:  $+0.9\text{‰} \div +3\text{‰}$  (Clayton & Mayeda 1999)
- CIs:  $+1.7\text{‰}$  (Leshin et al. 1997)
- CMs:  $> -0.7\text{‰}$  (Benedix et al. 2003)
- up to  $+20\text{‰}$  (Young et al. 1999; Young 2001)



# Sources of water on asteroids: Insights from H, N & O isotopes

**Fig. 2.** Comparison of the estimated hydrogen isotopic compositions of water in various chondrite groups with those measured in Oort cloud and Jupiter family comets (JFC), and Saturn's icy moon Enceladus. See (3) for details and sources.



- variable amounts of the outer solar system water re-equilibrated at high temperature during chondrule formation with nebular  $H_2$  acquiring  $\sim$ solar H-isotopic compositions
- expected inverse correlation between chondrule abundances in chondrites & their initial water D/H values

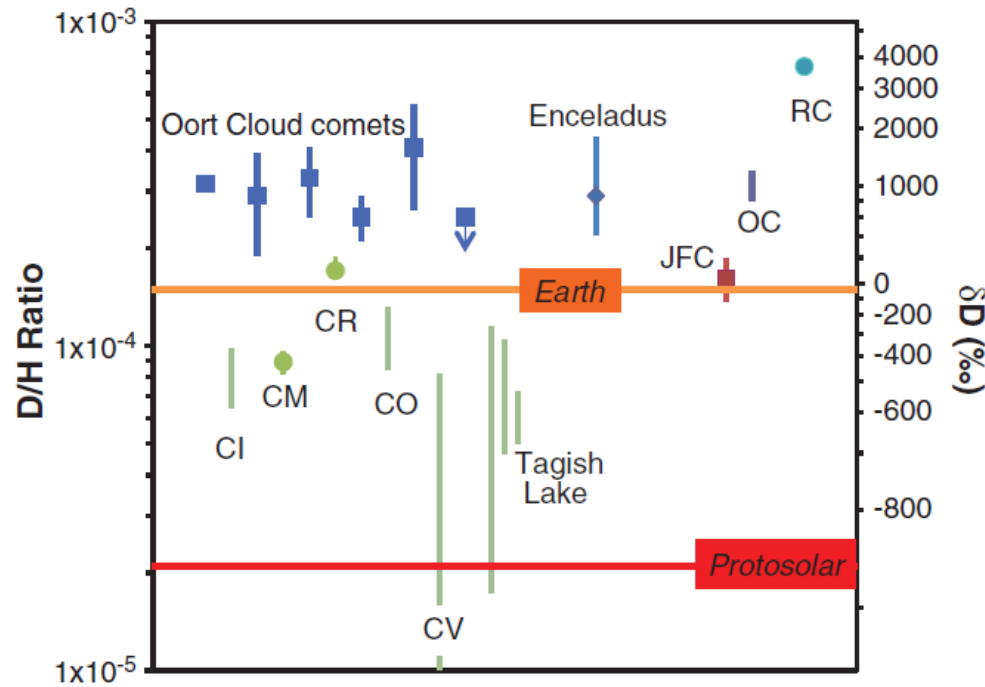
chd abundances:  $OC \approx RC > CV \approx CO \approx CR > CM > \text{Tagish Lake} > CI$

initial water D/H:  $RC > OC > CR > CV \approx CO \approx CM \approx \text{Tagish Lake} \approx CI$



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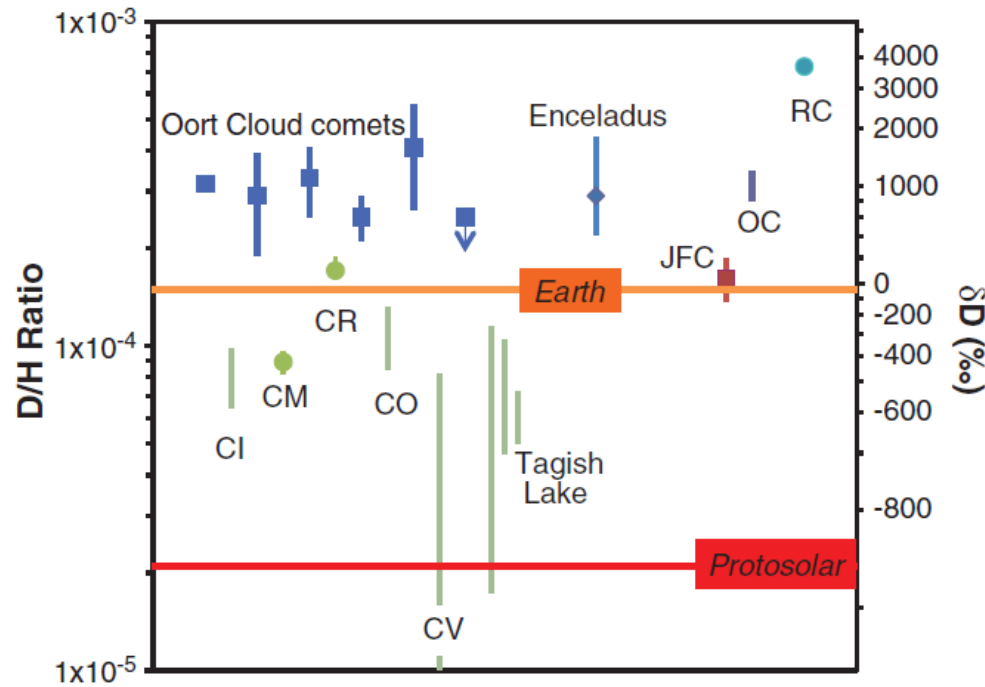
- isotopically solar-like water re-equilibrated with nebular  $H_2$  sunward of the snow line & subsequently migrated out to the chondrite-formation region beyond the snow line, e.g., via the cold finger effect

- water in OCs & RCs would then be essentially pure outer solar system ice
- water in OCs & RCs should be more  $^{16}O$ -depleted than in CCs (need to check)

*“Thus, at present there is inconsistency between the ice influx model & our results, unless, perhaps, the major ice influx occurred long before chondrite formation & the ice that chondrites accreted formed locally (with smaller isotopic anomalies) because transport in the disk had become less efficient”*

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